

# Mass Spectrometric Investigations of the Atmospheres of Giant Planets

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# Science Objectives

Scientific Objective <sup>1</sup>	Measurement Objective <sup>1</sup>	Measurement Requirements <sup>1</sup>
Constrain our models and understanding of Saturn's formation and subsequent evolution in situ measurement of elemental constituents and isotopic abundances for comparison with elemental compositions observed in the Sun, meteorites, and other planetary atmospheres	Determine fractional abundance of helium relative to H <sub>2</sub>	H <sub>2</sub> /He ratio to an accuracy of 1% <sup>2</sup>
	Determine atmospheric abundance ratios of C, H, O, N, and noble gases	C/H, N/H, O/H to a precision of ±10% or better Ne/He, Ar/He, Kr/He, Xe/He to a precision of ±30% or better
	Determine isotopic ratios of C, H, O, N, and noble gases	<sup>13</sup> C/ <sup>12</sup> C, <sup>18</sup> O/ <sup>16</sup> O to ±1% or better D/H, <sup>15</sup> N/ <sup>14</sup> N in major molecular species to ±1% to 5% Isotopic ratios of He, Ne, Ar, Kr, Xe to ±5% to ±10% or better <sup>3</sup>
Determine the strength of vertical mixing in Saturn's atmosphere	Determine the vertical abundance profiles of CH <sub>4</sub> , NH <sub>3</sub> , PH <sub>3</sub> , and H <sub>2</sub> S	Measure abundances with uncertainty of a factor of 2 or less <sup>4</sup>
Constrain models of cloud formation and structure	Determine abundances of condensable species below cloud base	e.g., H <sub>2</sub> O, NH <sub>3</sub> , and H <sub>2</sub> S to a factor of 2 or better <sup>4</sup>

<sup>1</sup>from *A Strategy for the Exploration of the Outer Planets*, 1986-1996 (NRC, 1986)

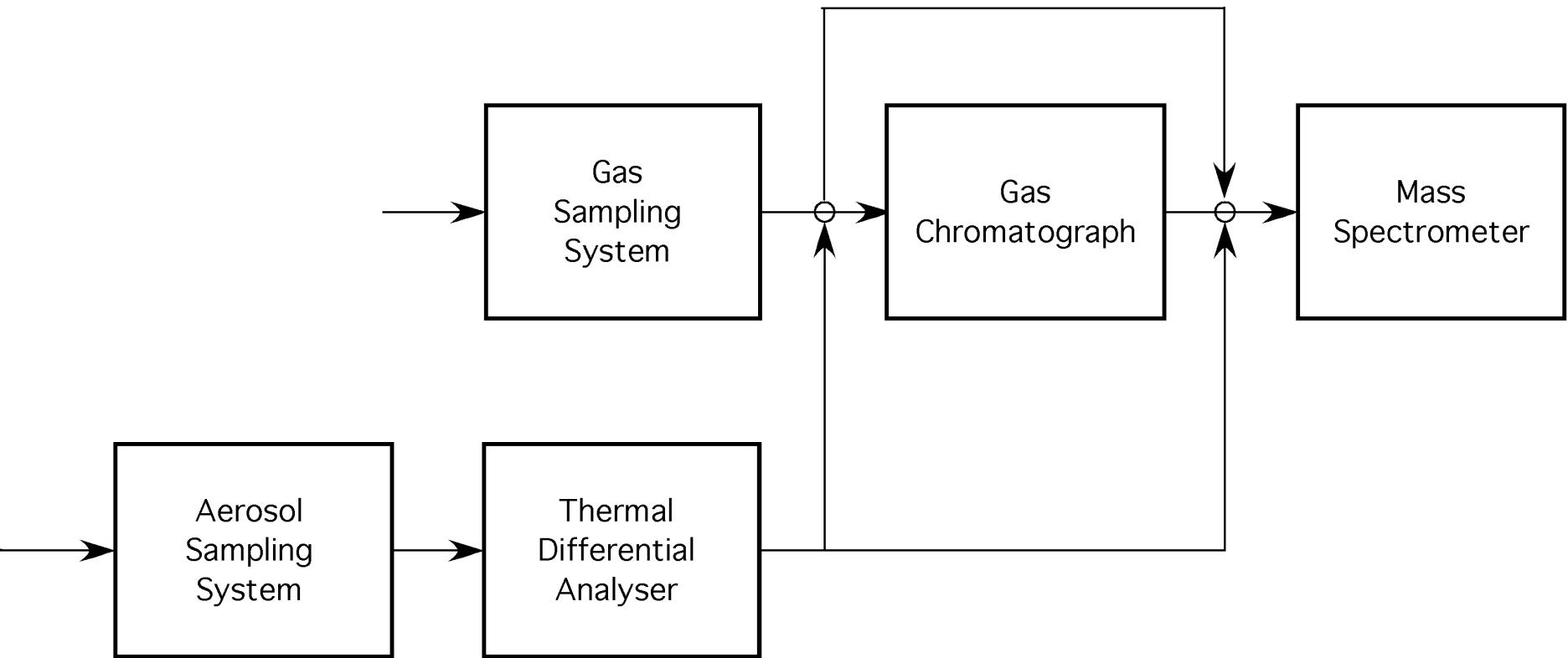
<sup>2</sup>RFI requires H/He to 5%

<sup>3</sup> "... except for <sup>21</sup>Ne, <sup>78</sup>Kr, and <sup>124,126</sup>Xe, where this accuracy may not be attainable" (*A Strategy*, p.23).

<sup>4</sup>from *A Strategy* but not included in the RFI list

Note: The RFI calls for measurement of sulfur to ±10% precision. This requirement does not appear in the COMPLEX report, but can, like the enhanced requirement for neon at 1% of solar, be satisfied by MASPEX. The reason for inclusion of <sup>20</sup>Ne/<sup>22</sup>Ne in the "Baseline Enhancements" is not clear. The <sup>20</sup>Ne/<sup>22</sup>Ne example is not used in the COMPLEX report (p. 23) to argue for "more precise or additional measurements," but to illustrate why a 10% error is better than a 30% error. An accuracy of 5-10% is already identified in the report (and the RFI) as the requirement for the noble gas isotopes (see above).

# Instrument Overview



# Galileo Probe Mass Spectrometer Experiment

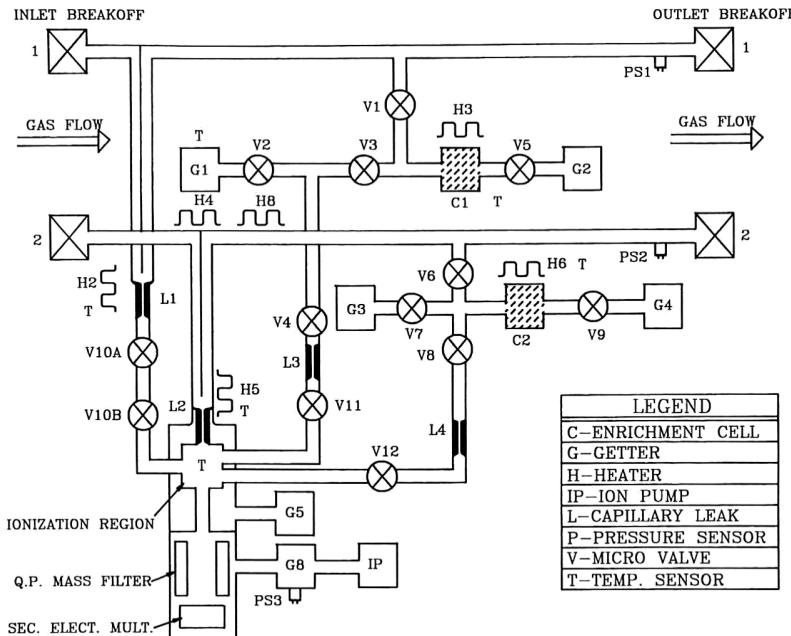


Fig. 1. Schematic of the gas inlet system and connection to the mass spectrometer sensor. Two parallel gas inlet/outlet systems are employed to provide gas samples to the direct leaks, L1 and L2, and to the two sample enrichment systems centered around C1 and C2.

H.B. Niemann, D.N. Harbold, S.K. Atreya, G.R. Cargnan, D.M. Hunten, T.C. Owen, Galileo Probe Mass spectrometer Experiment, Sp. Sci. Rev. 60 (1992) 111–142.

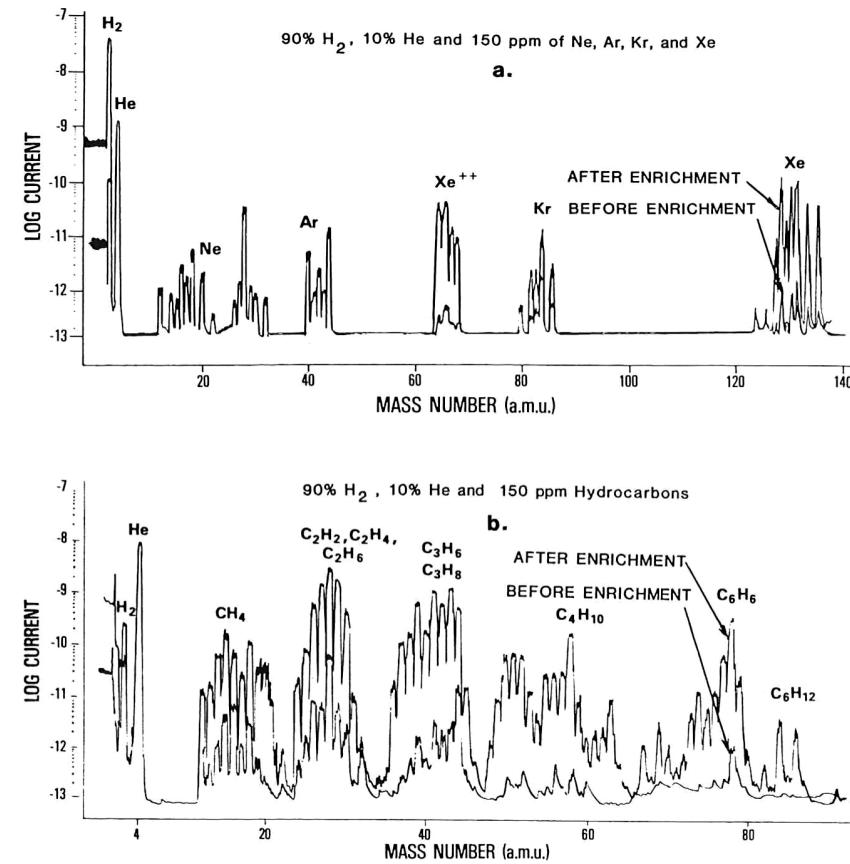
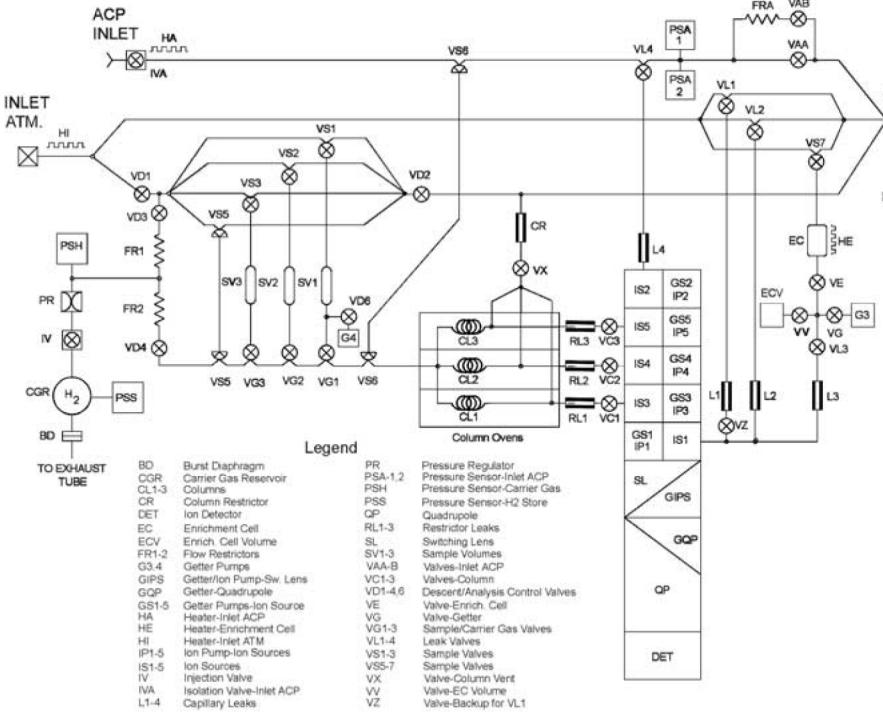


Fig. 2. Mass spectra showing enrichment obtained when gas processing is used to remove the major constituent H<sub>2</sub> from a 90% H<sub>2</sub> and 10% He mixture containing constituents each of 150 ppmv. (a) Rare gas enrichment. Note the substantial enrichment of Xe obtained. Expected Xe ratios are marginally measurable by direct analysis. (b) Hydrocarbon enrichment. There is a substantial enrichment of the C<sub>3</sub>–C<sub>4</sub> hydrocarbons with a somewhat less enrichment for methane. For illustration the spectra were recorded analog with laboratory recording equipment.

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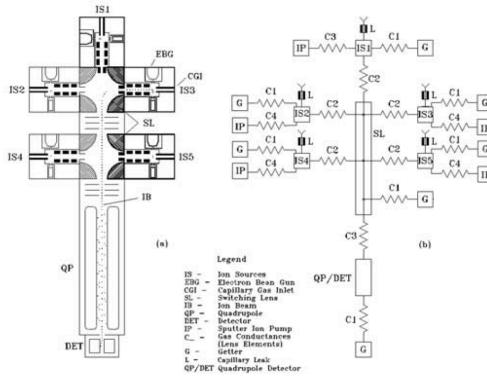
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# The Gas Chromatograph Mass Spectrometer for the Huygens Probe

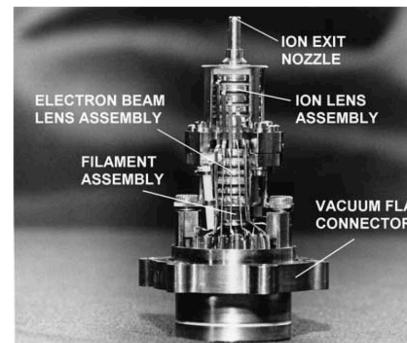


*Figure 3.* Schematic of the Gas Chromatograph Mass Spectrometer. Details of the Aerosol Collect Pyrolyser are shown in an accompanying paper in this volume.

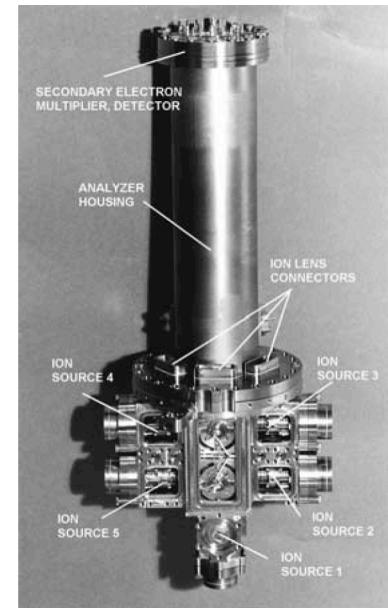
H.B. Niemann, et al., The Gas Chromatograph Mass Spectrometer for the Huygens Probe, Sp. Sci. Rev. 104 (2002) 553–591.



*Figure 5.* Illustrations of ion source configuration and schematic of differential vacuum pumping system.



*Figure 6.* An ion source showing the electron and ion focusing. The overall height is 63 mm.

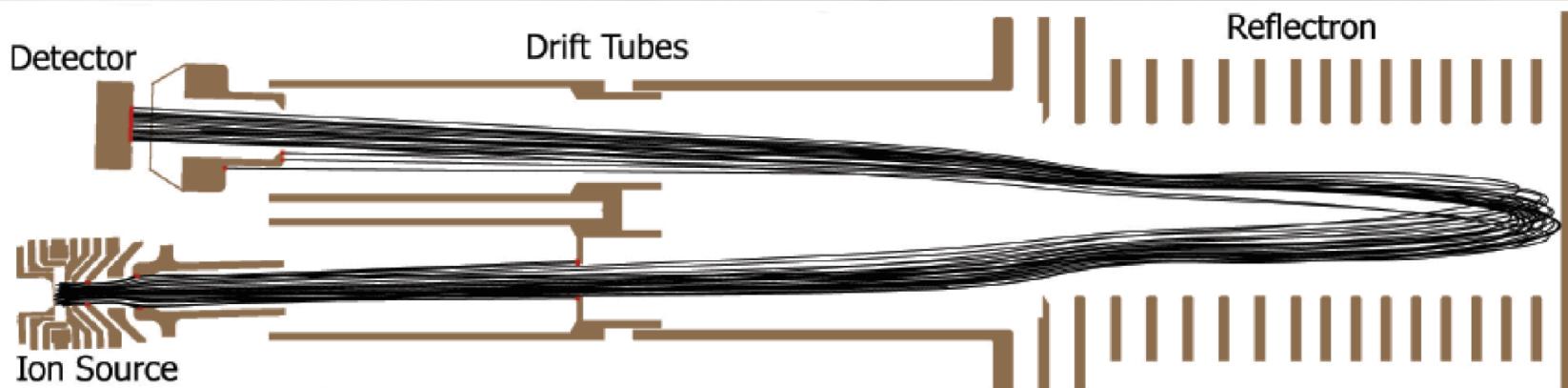
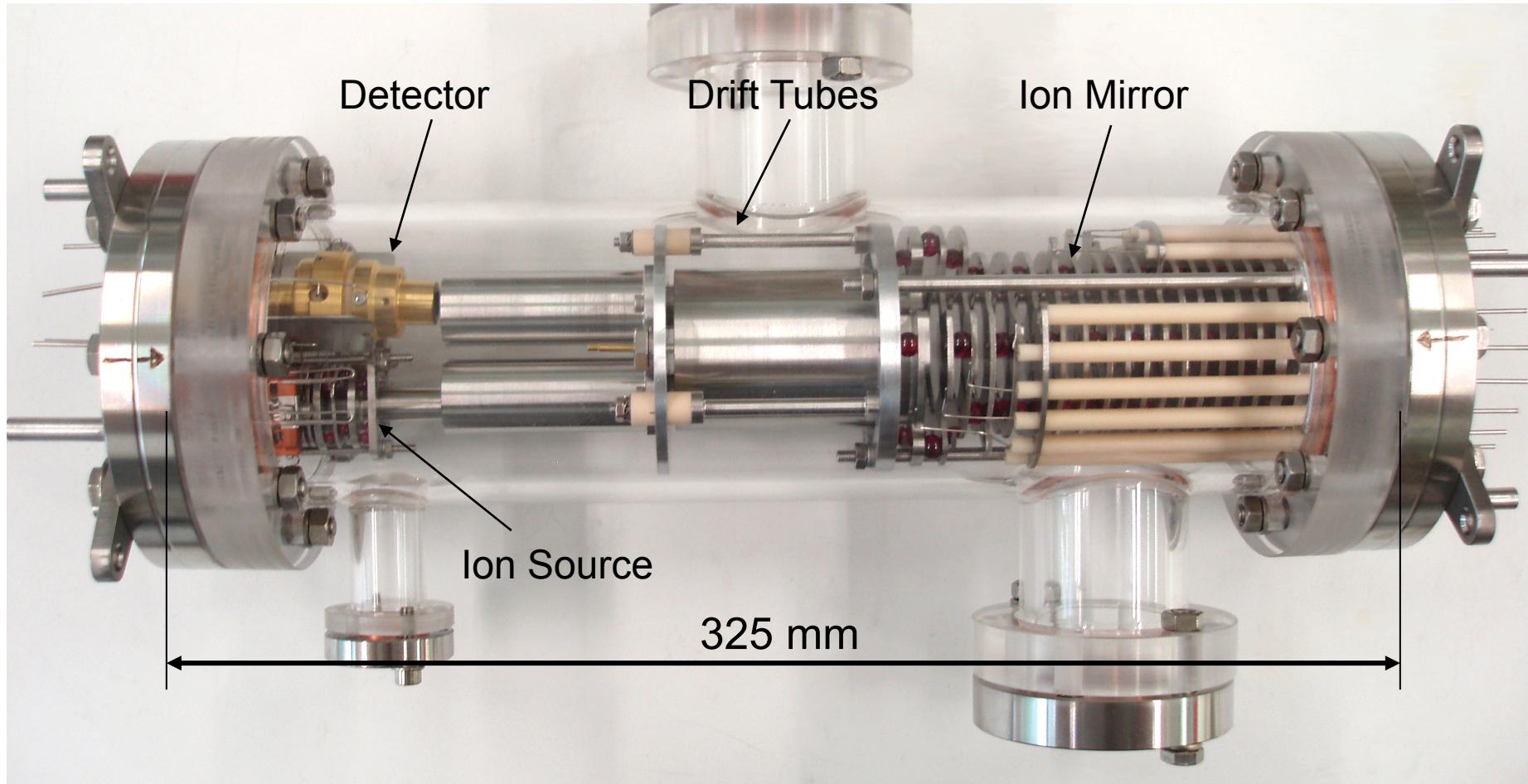


<sup>si</sup> Figure 7. The ion source system, showing the individual ion sources and the switching lenses in the partially assembled ion source housing before installation of the getter housings and the gas inlet manifolds. The overall length is 357 mm.

# Possible Future Developments

- ❖ Increase in sensitivity
- ❖ Increased mass range
- ❖ Increase in mass resolution
- ❖ Reduced complexity
- ❖ Reduced instrument resources

# Increase in Sensitivity



# P-BACE on the MEAP platform



Background image: NASA, <http://visibleearth.nasa.gov/>

- ❖ Balloon provided by Esrange Space Center to test beyond line of sight flight
- ❖ Semicircular flight following the summer polar vortex
- ❖ Launched from Esrange, Sweden, on 28 June 2008
- ❖ Altitude 33 ... 38 km
- ❖ Landed in Canada, near Umingmaktok, on 3 July 2008.
- ❖ Recorded ~ 4500 mass spectra in stratosphere

Species	Measurement 100 spectra (20080701/ 18:50 – 20:40)	Meas. Error [%]	Literature value	Remarks
$^{36}\text{Ar}/^{38}\text{Ar}$	5.313	7	5.35	
$^{36}\text{Ar}/\text{S}_{\text{tot}}$	28 ppm	7	32 ppm	
$^{36}\text{Ar}/^{40}\text{Ar}$	$7.2 \cdot 10^{-4}$	7	$3.38 \cdot 10^{-3}$	Discrimination of small signals (ADC-card)
$^{21}\text{Ne}/\text{S}_{\text{tot}}$	0.053 ppm	21	0.049 ppm	Possible interference with $\text{H}_2\text{F}^+$
$^{20}\text{Ne}/^{22}\text{Ne}$	40	80	9.782	Interferences with $^{40}\text{Ar}^{++}$ and $\text{CO}_2^{++}$
$^{78}\text{Kr}/^{84}\text{Kr}$	0.010	50	0.006	
$^{80}\text{Kr}/^{84}\text{Kr}$	0.040	50	0.040	
$^{82}\text{Kr}/^{84}\text{Kr}$	0.200	11	0.203	
$^{83}\text{Kr}/^{84}\text{Kr}$	0.200	11	0.202	
$^{86}\text{Kr}/^{84}\text{Kr}$	0.300	7	0.304	
$\text{Kr}_{\text{tot}}/\text{S}_{\text{tot}}$	1.15 ppm	6	1.14 ppm	
$^{129}\text{Xe}/^{132}\text{Xe}$	0.9615	14	0.9833	
$^{131}\text{Xe}/^{132}\text{Xe}$	0.8846	14	0.7877	
$^{134}\text{Xe}/^{132}\text{Xe}$	0.3462	22	0.3882	
$^{136}\text{Xe}/^{132}\text{Xe}$	0.3077	22	0.3299	
$\text{Xe}_{\text{tot}}/\text{S}_{\text{tot}}$	0.3 ppm	8	0.087 ppm	Consideration of the large cross section for Xe gives a ratio of ~0.08 ppm
$^{196}\text{Hg}/^{202}\text{Hg}$	0.020	82	0.005	
$^{198}\text{Hg}/^{202}\text{Hg}$	0.367	14	0.334	
$^{199}\text{Hg}/^{202}\text{Hg}$	0.600	14	0.565	
$^{200}\text{Hg}/^{202}\text{Hg}$	0.800	14	0.774	
$^{201}\text{Hg}/^{202}\text{Hg}$	0.467	14	0.441	
$^{204}\text{Hg}/^{202}\text{Hg}$	0.233	14	0.230	
$\text{Hg}_{\text{tot}}/\text{S}_{\text{tot}}$	1.72 ppm	6	--	Source unknown (gondola)
$\text{O}_3/\text{S}_{\text{tot}}$	0.59 ppm	21	8 ppm	Pronounced fragmentation
H/D (H <sub>2</sub> /HD)	8800	14	8694	
$^{23}\text{Na}^+/\text{S}_{\text{tot}}$	0.99 ppm	11	?	Probably oceanic NaCl
$^{40}\text{Ca}^{16}\text{O}_2^+/\text{S}_{\text{tot}}$	0.07 ppm	21	?	Contamination or meteorite material

P. Wurz, D. Abplanalp, M. Tulej, M. Iakovleva, V.A. Fernandes, A. Chumikov, and G. Managadze, "In Situ Mass Spectrometric Analysis in Planetary Science," Sol. Sys. Res. (2012) in press.

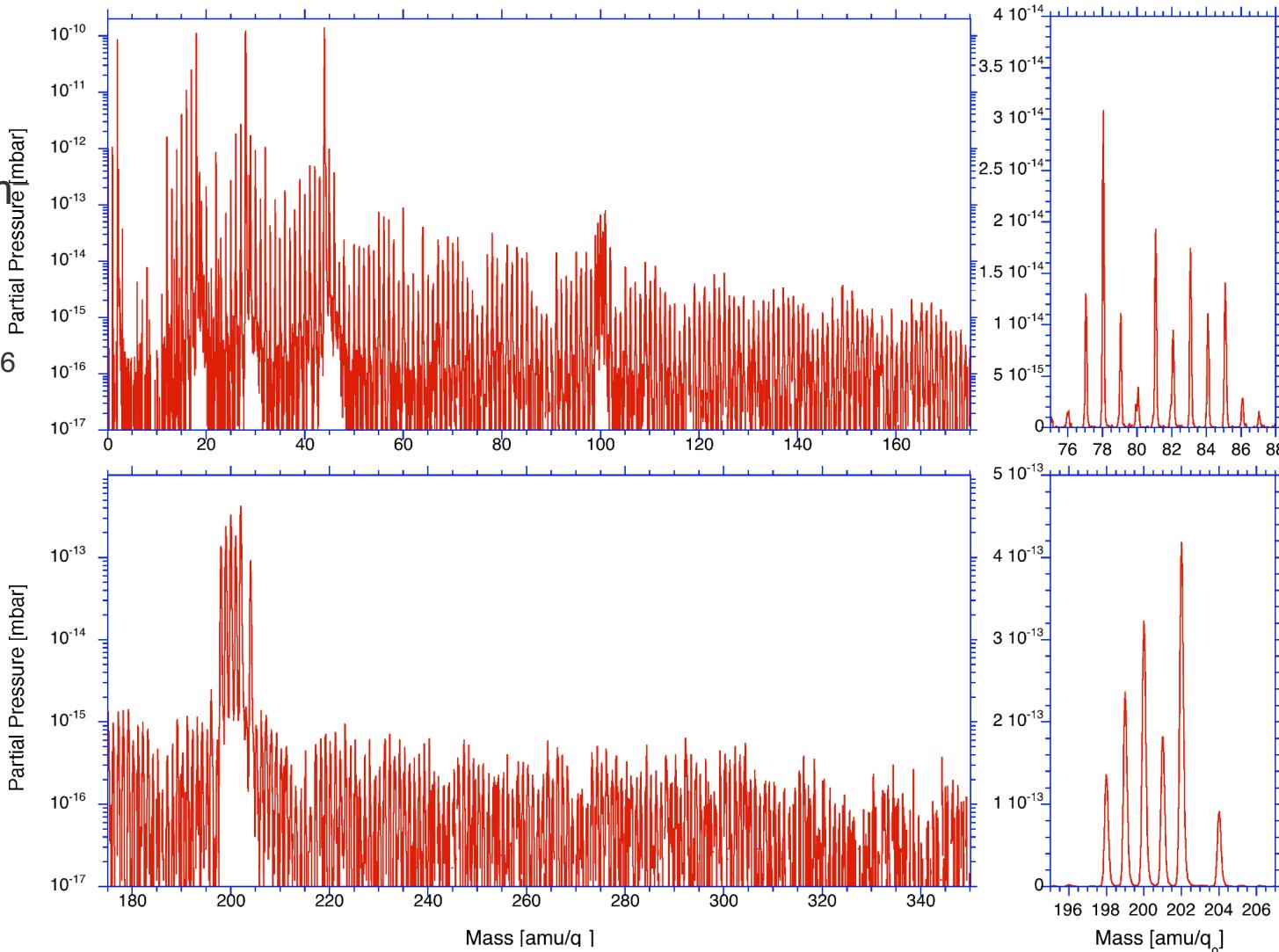
# Neutral Gas Mass Spectrometer Luna-Resurs / Luna-Glob

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## Sensitivity

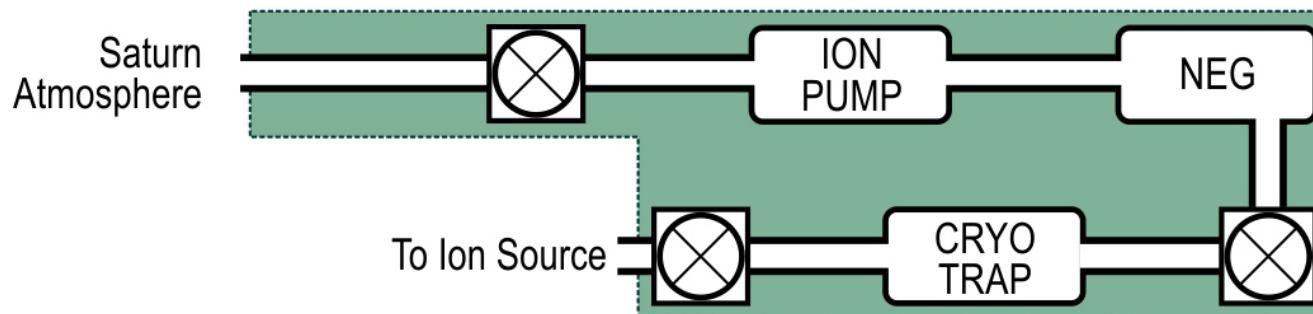
- ❖ Detection threshold:  $\approx 1 \text{ cm}^3$  for 10 second integration
- ❖ Detection threshold:  $\approx 10^{-16} \text{ mbar}$  for 10 second integration



# Noble Gas Sample Concentration: Cryotrap

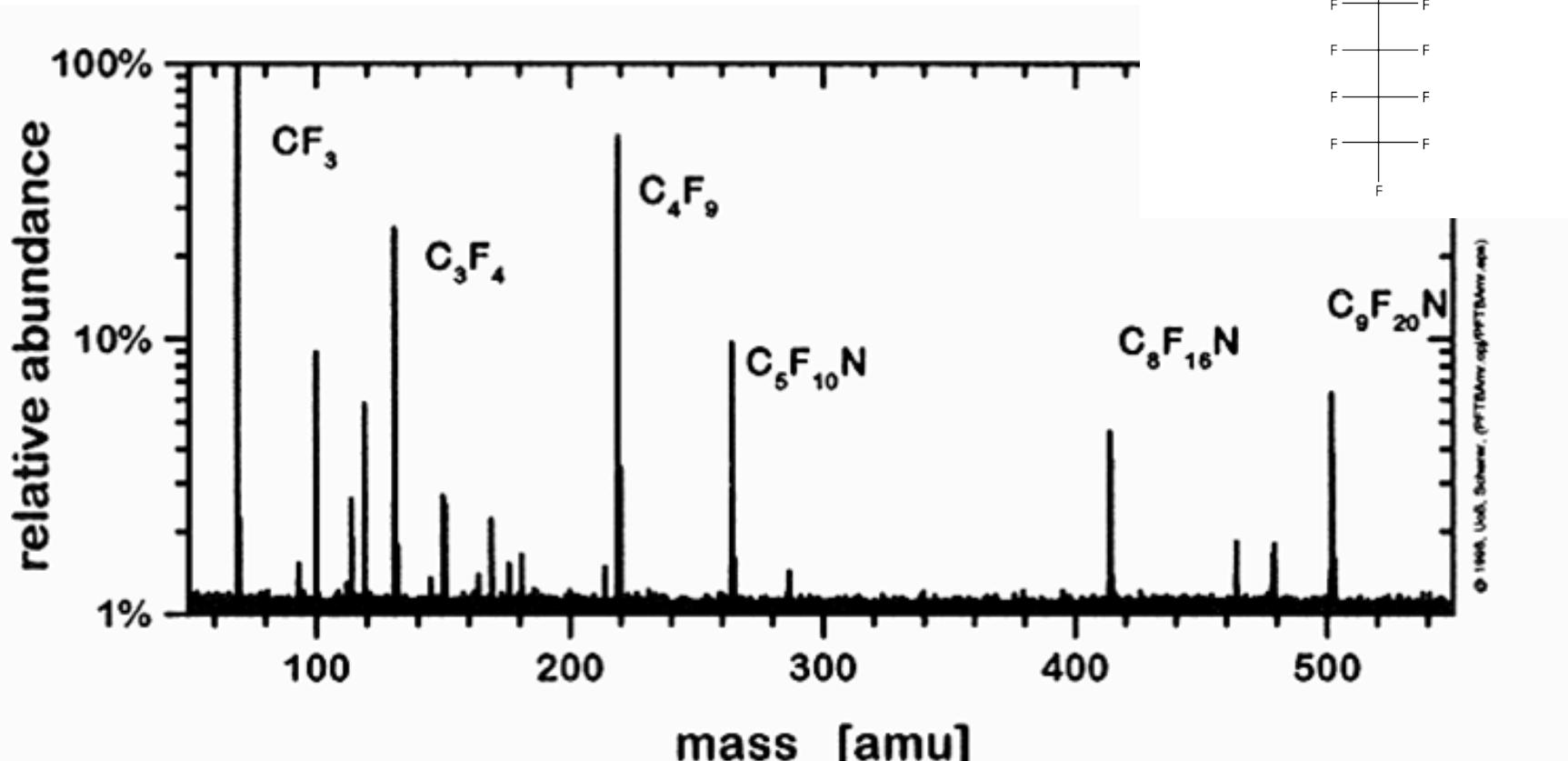
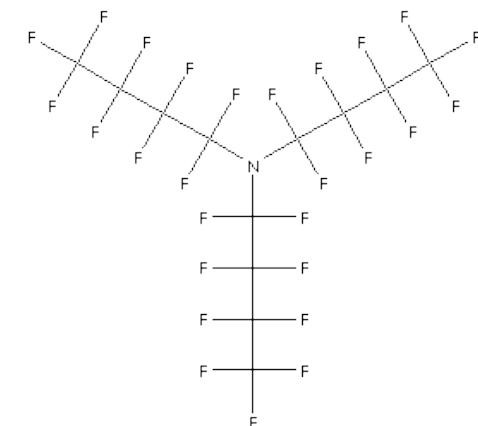
## ❖ Sample concentration

- Even though a TOF-MS is over 1000 times more sensitive than Cassini INMS (10x from ion source efficiency and 100 x from better duty cycle), calculations show that it would take longer than the probe descent time to provide the counting statistics required for the isotopes of Kr and Xe with sufficient accuracy
- The noble gas enhancement can be achieved by using a combination of a cryotrap, ion pump, and non-evaporable getter (NEG: SAES 172).
- The NEG removes all constituents except methane and the noble gases.
- The cryotrap traps the products of the NEG process, except for helium and some neon.
- The ion pump then operates to pump away the helium, which is the second highest source of gas, thus enhancing the remaining noble gases ~ 200 times.
- Helium and neon are measured using a separate mode.



# Increased Mass Range

# RTOF / ROSINA / Rosetta Mass Range



RTOF mass spectrum (prototype) of the calibration compound heptacosafaurotributylamine, a compound to demonstrate mass range. Mass range is unlimited by sensor itself, data acquisition memory however limits the mass range that can be covered. In case of RTOF, the mass range is up to 1000 amu.

S. Scherer, K. Altwegg, H. Balsiger, J. Fischer, A. Jäckel, A. Korth, M. Mildner, D. Piazza, H. Rème, and P. Wurz, "A novel principle for an ion mirror design in time-of-flight mass spectrometry," Int. Jou. Mass Spectr. 251 (2006) 73-81.

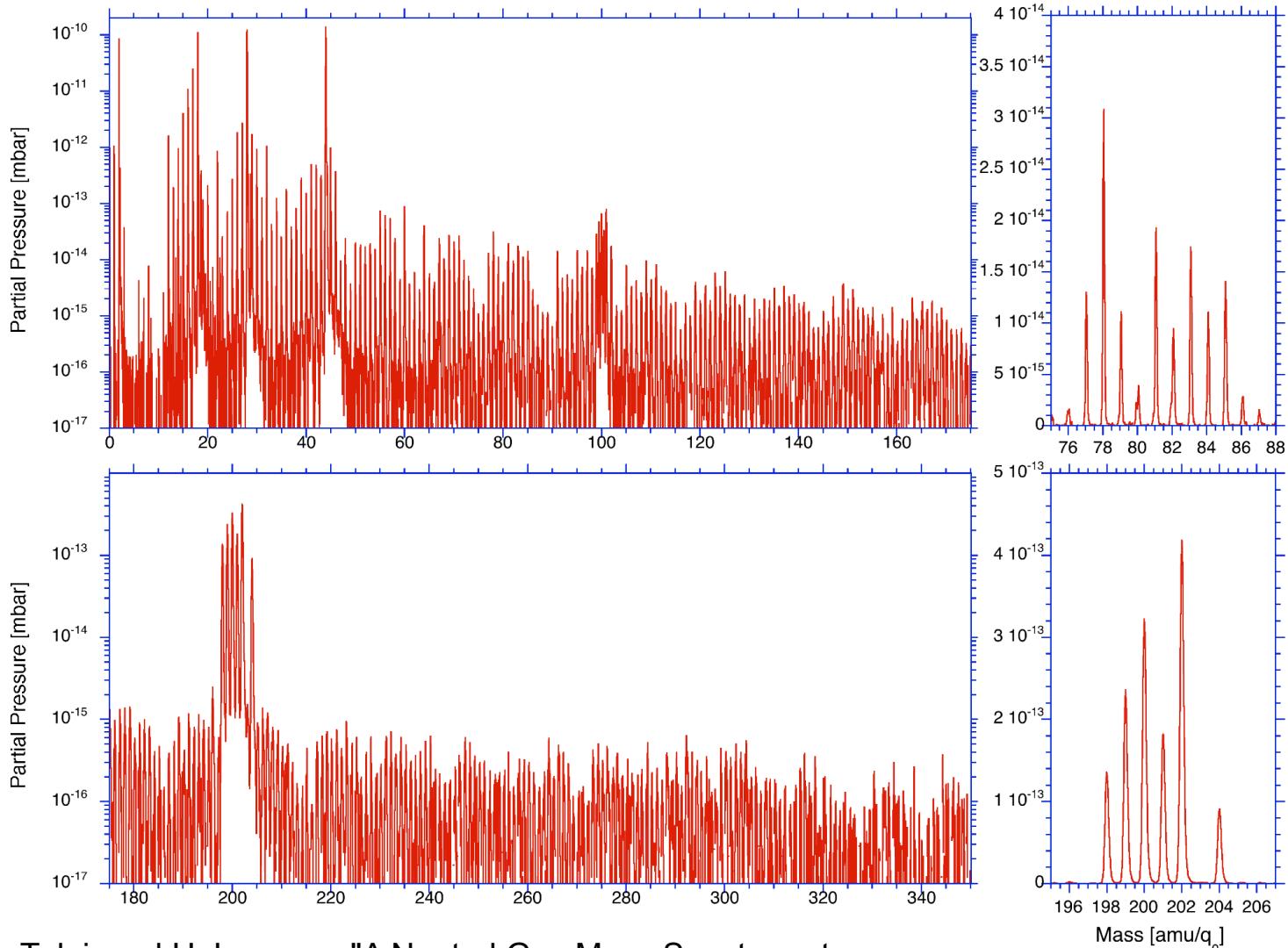
# Neutral Gas Mass Spectrometer Luna-Resurs / Luna-Glob

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## Mass range

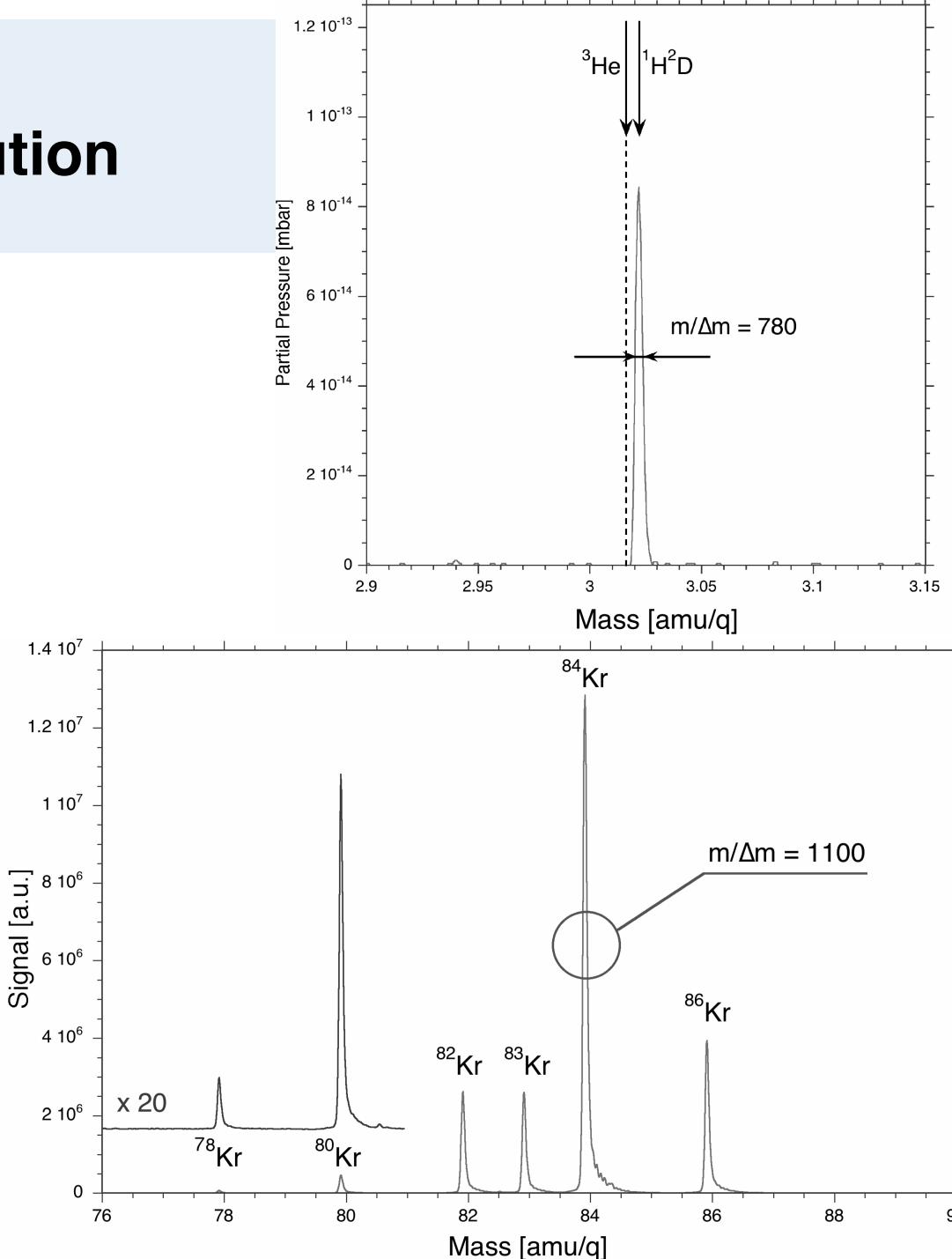
- ❖ 1 – 300 amu (low mass range mode)
- ❖ Can be increased in flight by command, if necessary, to 1000 amu (high mass range mode).
- ❖ Mass range is only limited by memory for accumulation



# Increased Mass Resolution

# Increased Mass Resolution

- ❖  $m/\Delta m \approx 100$  (50%)
  - there is separation of low-mass chemical species
- ❖  $m/\Delta m \approx 1000$  (50%)
  - there is a separation of peaks of different nominal mass (e.g., 325 amu versus 326 amu)
- ❖  $m/\Delta m \approx 10000$  (50%),
  - separation of peaks for nominally isobaric species (i.e., molecules of the same nominal mass differing in elemental composition), e.g., N<sub>2</sub> versus CO, both ~28 amu.
  - resolution of small (< 2500 amu) peptides of the same nominal mass differ by one amino acid (except for isomeric leucine and isoleucine)
- ❖  $m/\Delta m \approx 100000$  (50%),
  - there is a separation of peaks for nominally isobaric species, e.g. for complex carbon chemistry



P. Wurz, D. Abplanalp, M. Tulej, and H. Lammer,  
"A Neutral Gas Mass Spectrometer for the  
Investigation of Lunar Volatiles," *Planet. Sp. Science* (2012) in print.

# Rosetta / ROSINA Instruments



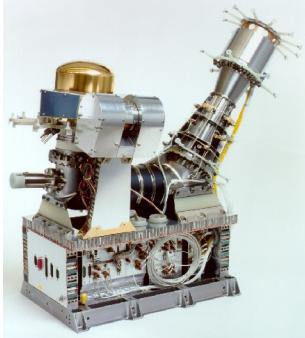
## COPS:

- Measures the total neutral particle density and the ram pressure (the cometary gas flux)
- Measures the total pressure down to densities of  $10^4 \text{ cm}^{-3}$  in 10s



## RTOF:

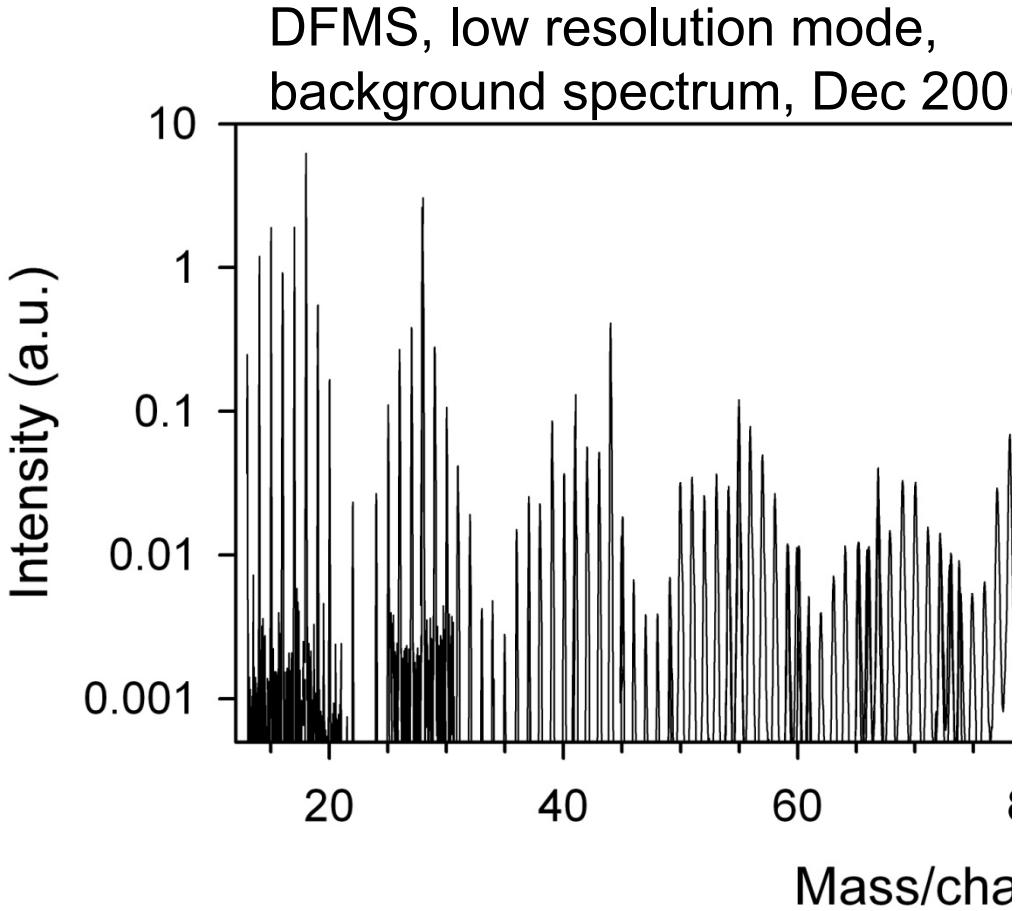
- Covers mass range from 1 to  $> 300 \text{ amu/e}$
- Mass resolution  $m/\Delta m > 3000$  (at 50% peak height)
- Detects particle densities of  $10^2 \text{ cm}^{-3}$  within 10–1000s for a complete spectrum



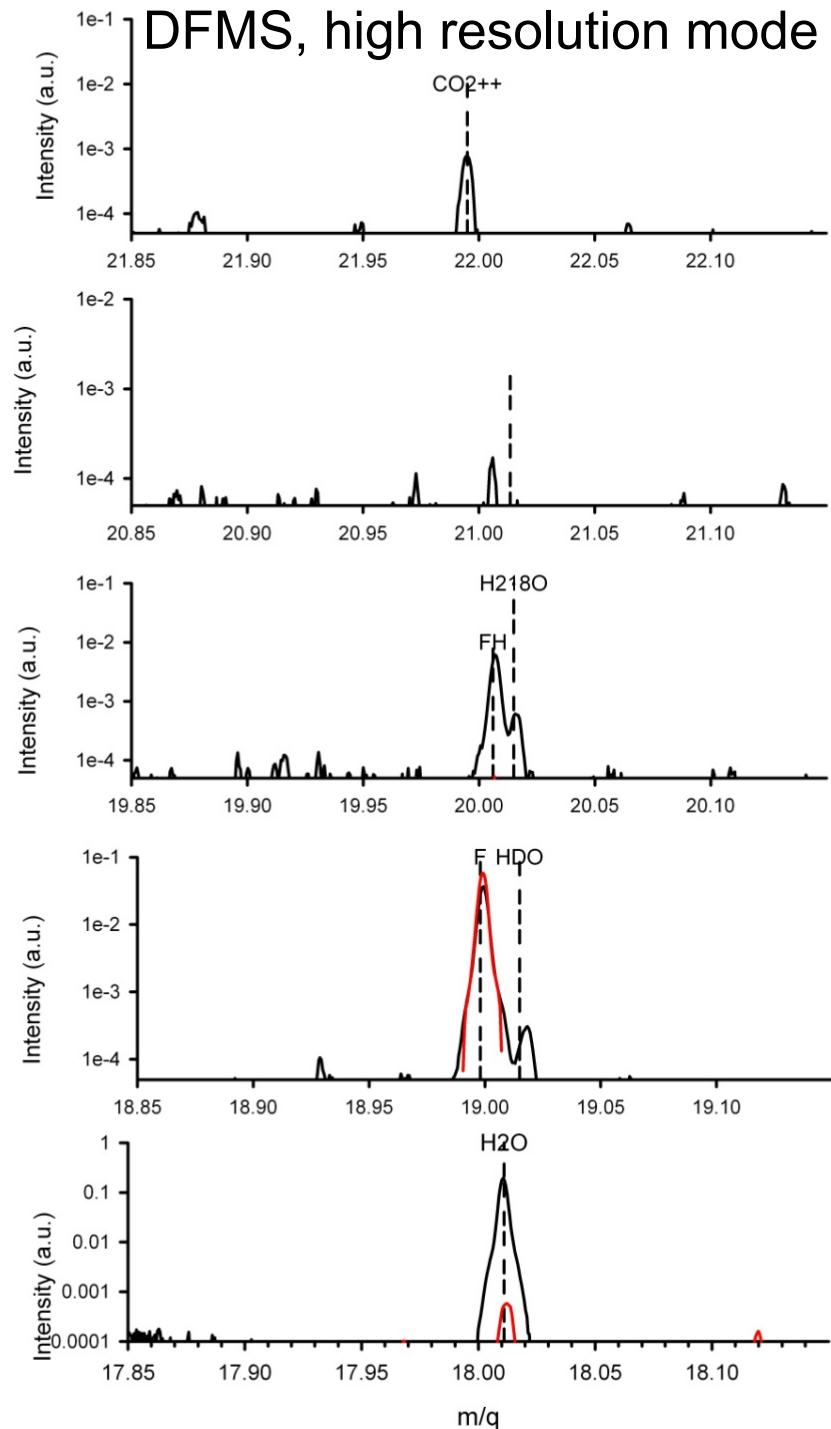
## DFMS:

- Covers mass range from 12 to 150 amu/e
- Mass resolution  $m/\Delta m > 9000$  (at 50% peak height)
- Detects particle densities of  $1 \text{ cm}^{-3}$  within 20 s for one mass line
- Complete mass spectrum  $\sim 20 \text{ min}$

# Rosetta / ROSINA: Active comet Double Focussing Mass Spectrometer

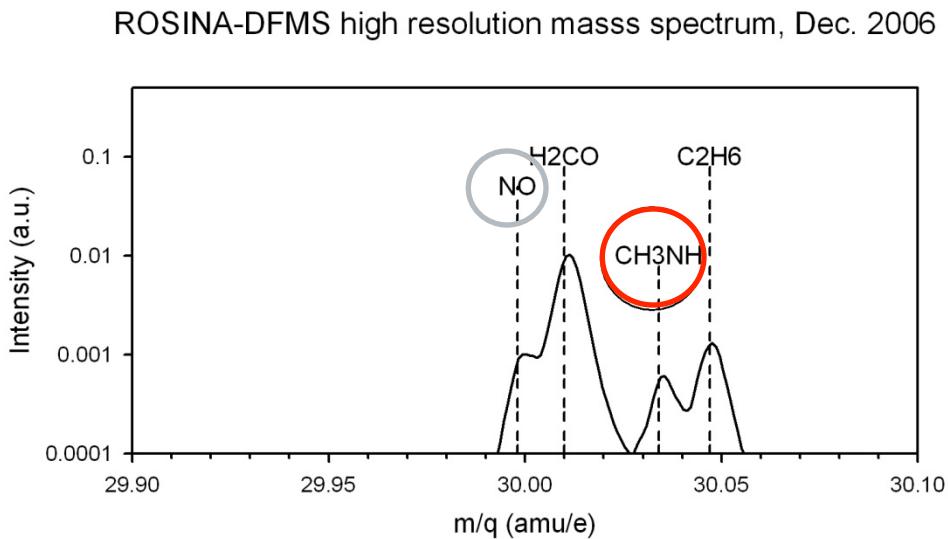


B. Schl  ppi, K. Altwegg, H. Balsiger, M. H  ssig, A. J  ckel, P. Berthelier, J. DeKeyser, H. R  me, and U. Mall, "The influence of tenuous atmospheres with in situ mass spectrometry," Jou. C.

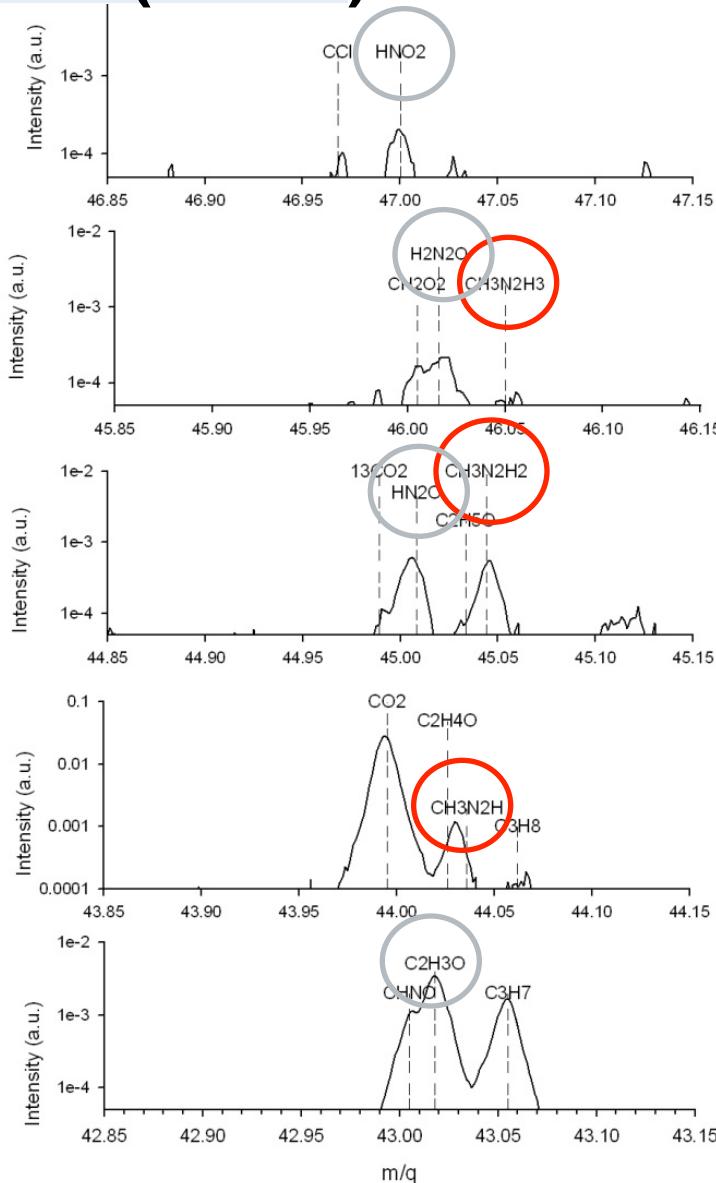


# Rosetta / ROSINA: Active checkout PC4 Double Focussing Mass Spectrometer (DFMS)

Traces of hydrazine  $\text{CH}_3\text{N}_2\text{H}_3$ , its oxidant  $\text{N}_2\text{O}_4$ , and products, 20 h after thruster firing



$$P_{\text{total}} \sim 4 \times 10^{-11} \text{ mbar}, P_{\text{hydrazine}} \sim 1 \times 10^{-12} \text{ mbar}$$



# Some identified molecules and fragments in the vicinity of Rosetta S/C

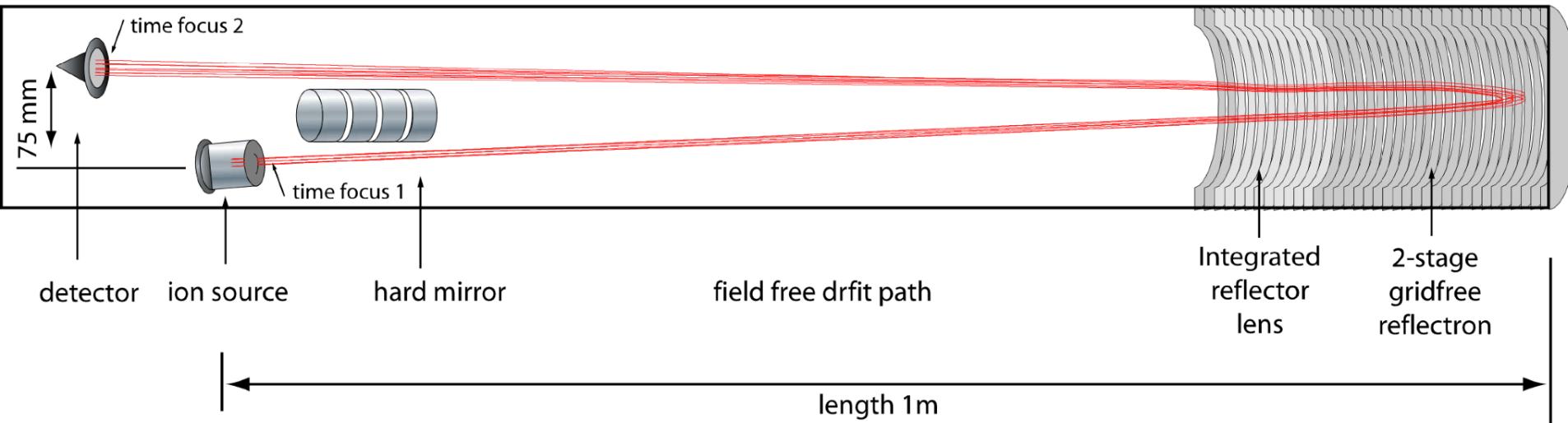
Carbohydrates				PAH		N-H	Hydrazine	C-N	Oxygen	N-O
C2	C3	C4	C5			N	CN		O	NO
CH	C <sub>2</sub> H	C <sub>3</sub> H	C <sub>4</sub> H	C <sub>5</sub> H	C <sub>6</sub> H	NH	CHN	C <sub>2</sub> H <sub>2</sub> N	OH	CNO
CH <sub>2</sub>	C <sub>2</sub> H <sub>2</sub>	C <sub>3</sub> H <sub>2</sub>	C <sub>4</sub> H <sub>2</sub>	C <sub>5</sub> H <sub>2</sub>	C <sub>6</sub> H <sub>2</sub>	NH <sub>2</sub>	CH <sub>2</sub> N	C <sub>2</sub> H <sub>3</sub> N	OH <sub>2</sub>	HCNO
CH <sub>3</sub>	C <sub>2</sub> H	C <sub>3</sub> H <sub>3</sub>	C <sub>4</sub> H <sub>3</sub>	C <sub>5</sub> H <sub>3</sub>	C <sub>6</sub> H <sub>3</sub>	NH <sub>3</sub>	CH <sub>3</sub> N	C <sub>2</sub> H <sub>4</sub> N	ODH	H <sub>6</sub> CNO
CH <sub>4</sub>	C <sub>2</sub> H <sub>4</sub>	C <sub>3</sub> H <sub>4</sub>	C <sub>4</sub> H <sub>4</sub>	C <sub>5</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>4</sub>	N <sub>2</sub>	CH <sub>3</sub> NH		<sup>18</sup> OH <sub>2</sub>	NO <sub>2</sub>
	C <sub>2</sub> H <sub>5</sub>	C <sub>3</sub> H <sub>5</sub>	C <sub>4</sub> H <sub>5</sub>	C <sub>5</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>		CH <sub>3</sub> NH <sub>2</sub>	C <sub>5</sub> H <sub>4</sub> N	O <sub>2</sub>	HNO <sub>2</sub>
	C <sub>2</sub> H <sub>6</sub>	C <sub>3</sub> H <sub>6</sub>	C <sub>4</sub> H <sub>6</sub>	C <sub>5</sub> H <sub>6</sub>		C <sub>7</sub> H <sub>6</sub>	CH <sub>3</sub> N <sub>2</sub> H	C <sub>5</sub> H <sub>5</sub> N		H <sub>4</sub> NO <sub>2</sub>
		C <sub>3</sub> H <sub>7</sub>	C <sub>4</sub> H <sub>7</sub>	C <sub>5</sub> H <sub>7</sub>		C <sub>7</sub> H <sub>7</sub>	CH <sub>3</sub> N <sub>2</sub> H <sub>2</sub>	C <sub>5</sub> H <sub>6</sub> N		CHNO <sub>2</sub>
		C <sub>3</sub> H <sub>8</sub>	C <sub>4</sub> H <sub>8</sub>	C <sub>5</sub> H <sub>8</sub>		C <sub>7</sub> H <sub>8</sub> Toluene	CH <sub>3</sub> N <sub>2</sub> H <sub>3</sub>	C <sub>5</sub> H <sub>7</sub> N		CH <sub>3</sub> NO <sub>2</sub>
			C <sub>4</sub> H <sub>9</sub>	C <sub>5</sub> H <sub>9</sub>		C <sub>8</sub> H <sub>10</sub>		C <sub>5</sub> H <sub>8</sub> N		CH <sub>4</sub> NO <sub>2</sub>
			C <sub>4</sub> H <sub>10</sub>	C <sub>5</sub> H <sub>10</sub>		C <sub>9</sub> H <sub>12</sub>				C <sub>2</sub> H <sub>6</sub> NO
				C <sub>5</sub> H <sub>11</sub>						H <sub>2</sub> N <sub>2</sub> O
				C <sub>5</sub> H <sub>12</sub>						C <sub>2</sub> N <sub>2</sub> O
										C <sub>2</sub> HN <sub>2</sub> O
										C <sub>2</sub> H <sub>2</sub> N <sub>2</sub> O
										C <sub>2</sub> H <sub>3</sub> N <sub>2</sub> O
										C <sub>2</sub> H <sub>5</sub> N <sub>2</sub> O
										C <sub>2</sub> H <sub>6</sub> N <sub>2</sub> O
										C <sub>2</sub> H <sub>7</sub> N <sub>2</sub> O
										C <sub>2</sub> H <sub>8</sub> N <sub>2</sub> O

Yellow: Possible fragments from Monomethylhydrazine and N<sub>2</sub>O<sub>4</sub>

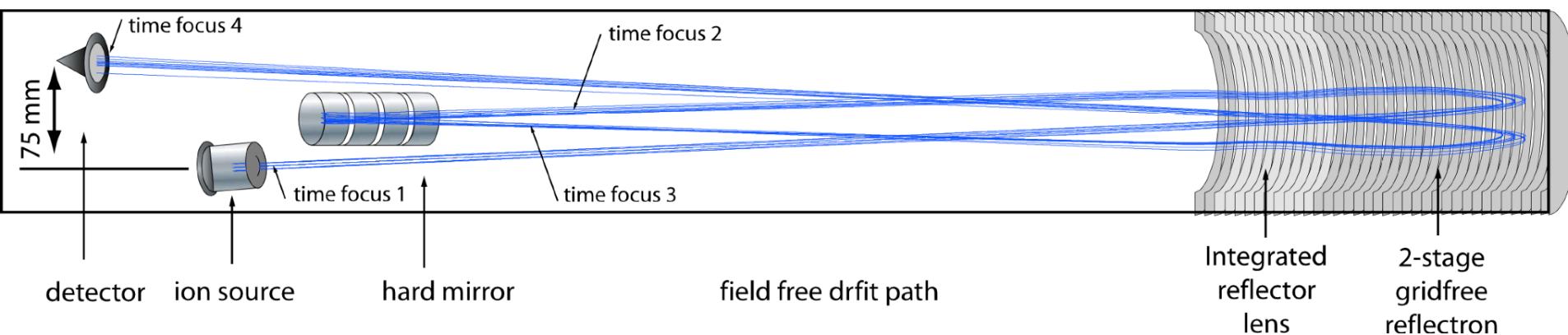
Green: Solvents

# RTOF / ROSINA / Rosetta

a) single reflection mode



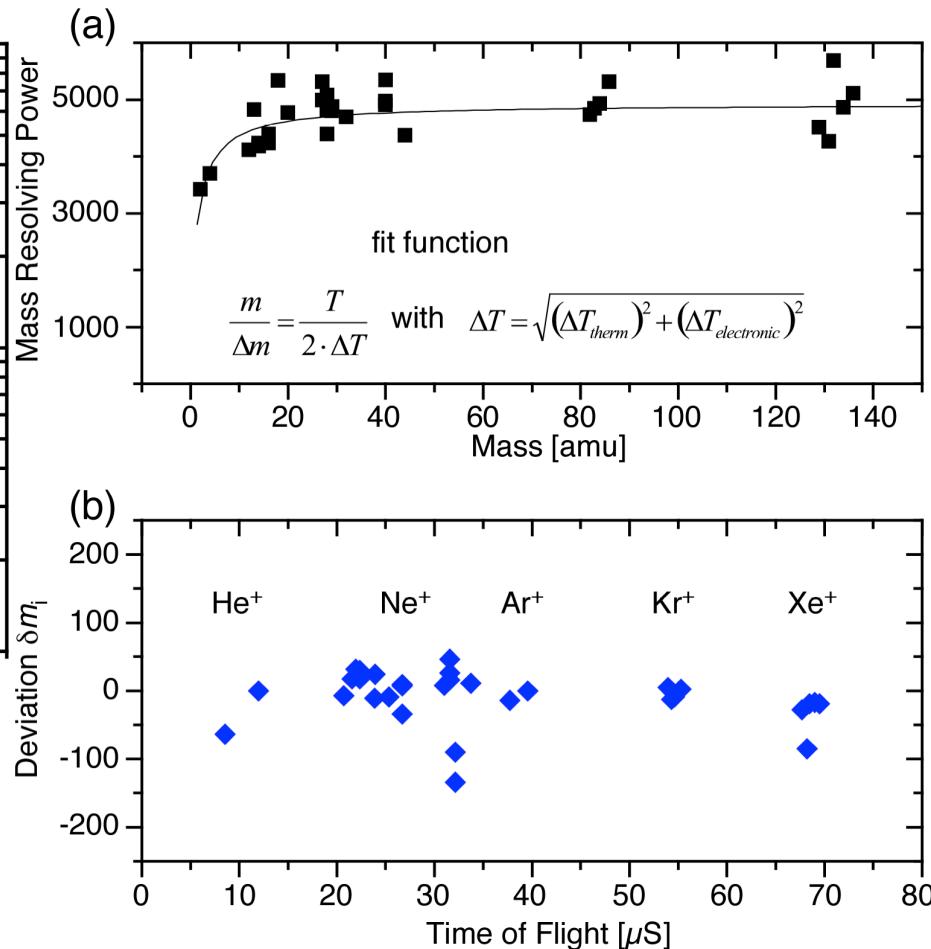
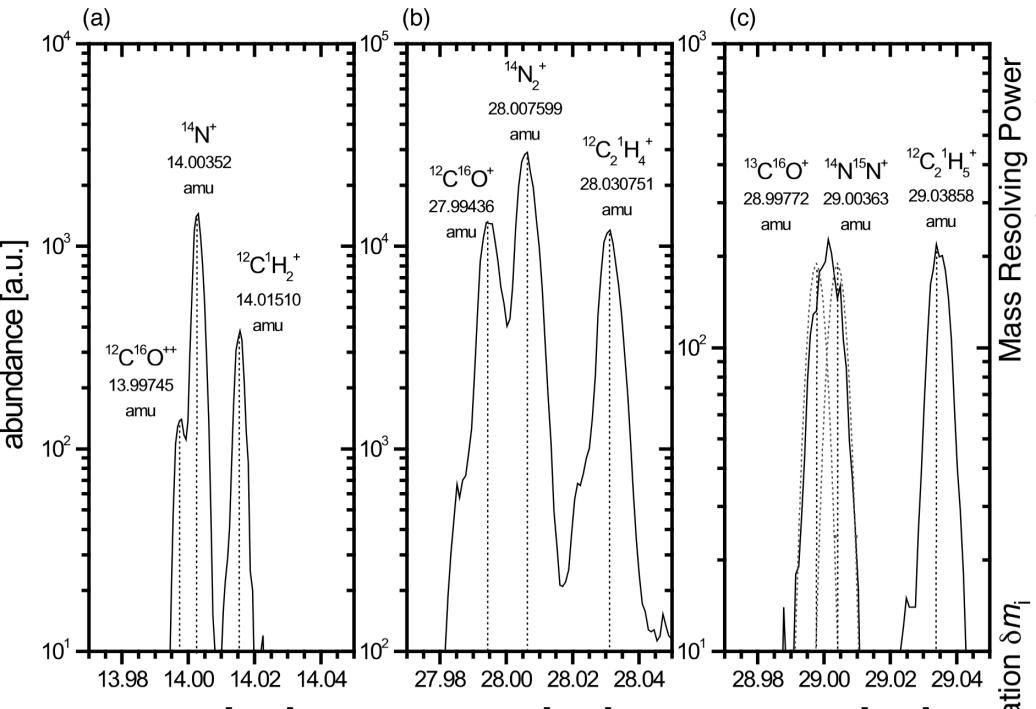
b) triple reflection mode



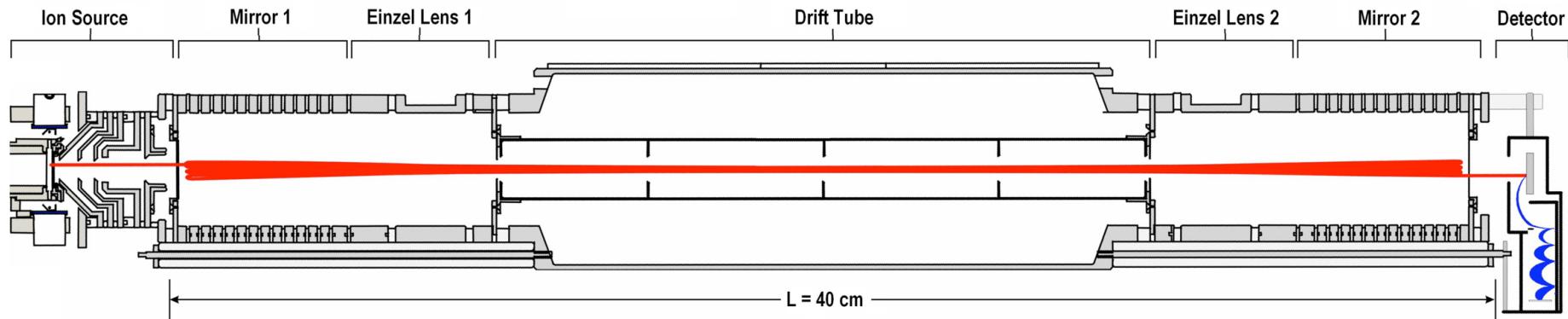
M. Hohl, P. Wurz, S. Scherer, K. Altwegg, and H. Balsiger, Int. J. Mass Spectr. 188 (1999), 189–197.

S. Scherer, K. Altwegg, H. Balsiger, J. Fischer, A. Jäckel, A. Korth, M. Mildner, D. Piazza, H. Rème, and P. Wurz, Int. Jou. Mass Spectr. 251 (2006) 73–81.

# Rosetta / ROSINA / RTOF

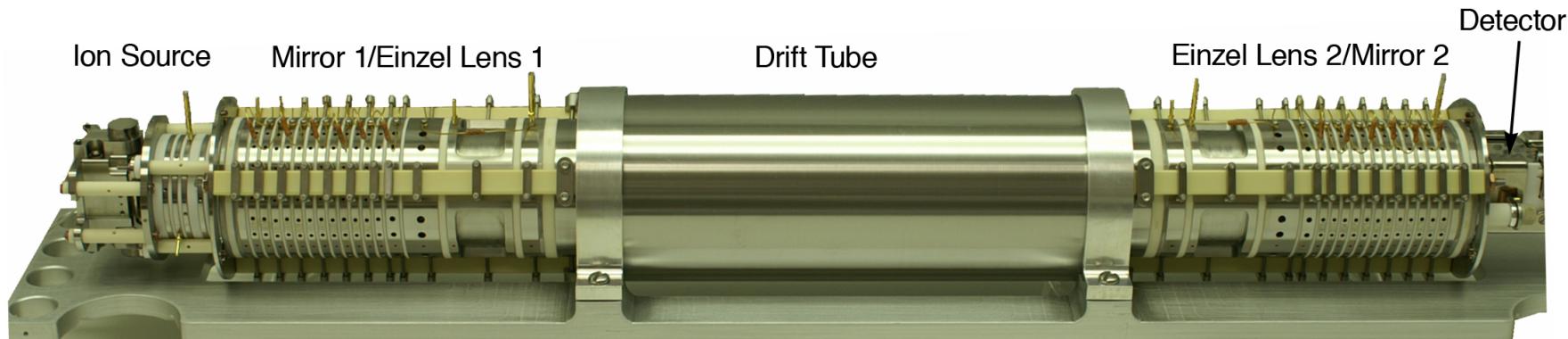


# Multi-Bounce time-of-flight mass spectrometer (MASPEX) Ion Optics Simulations and Prototype



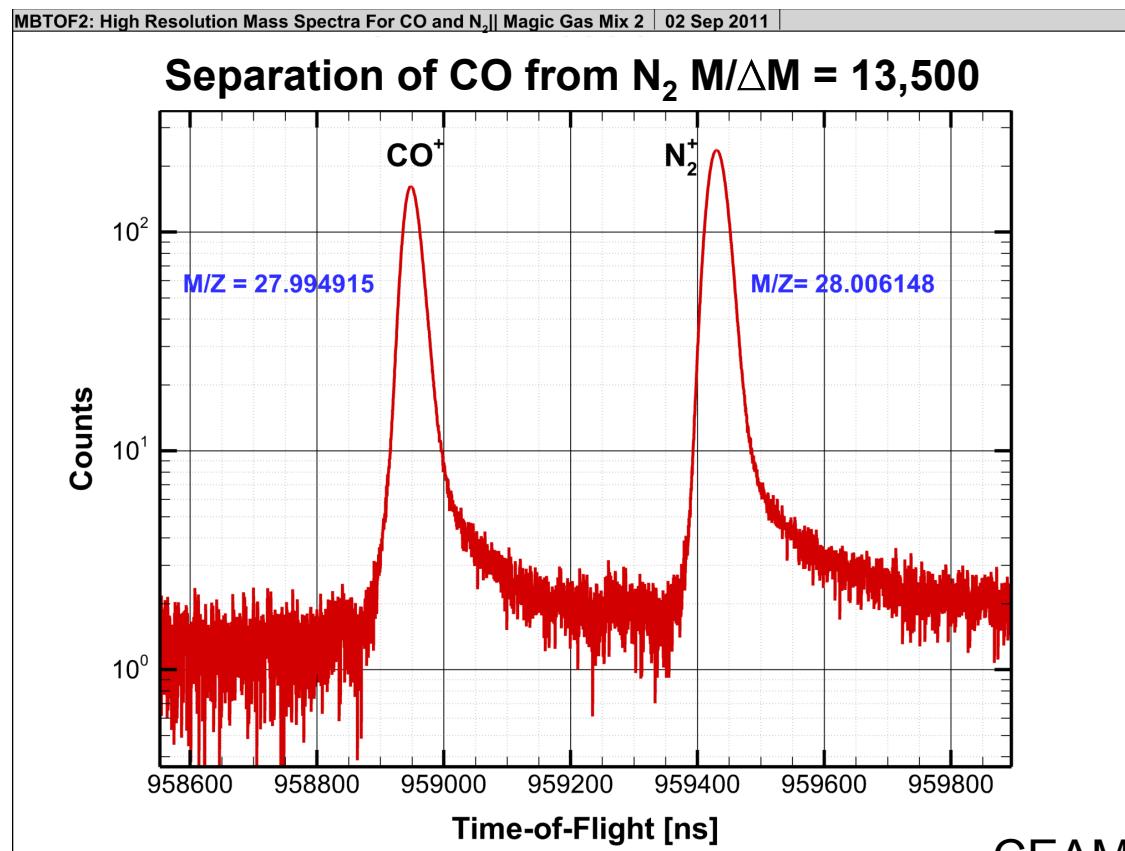
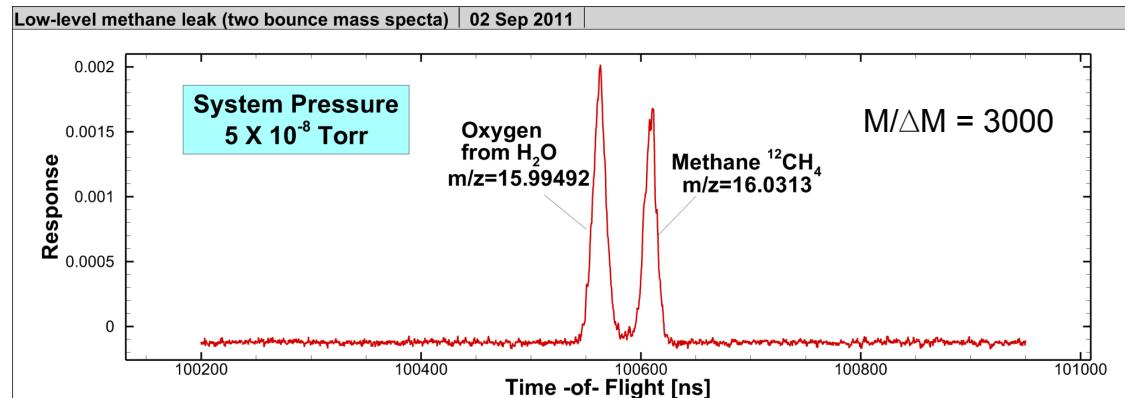
Numerical simulations of ion trajectories showing focusing of multi-bounce ion packets.

$$m/\Delta m(N) = NT_o/2(Dt+NdT)$$

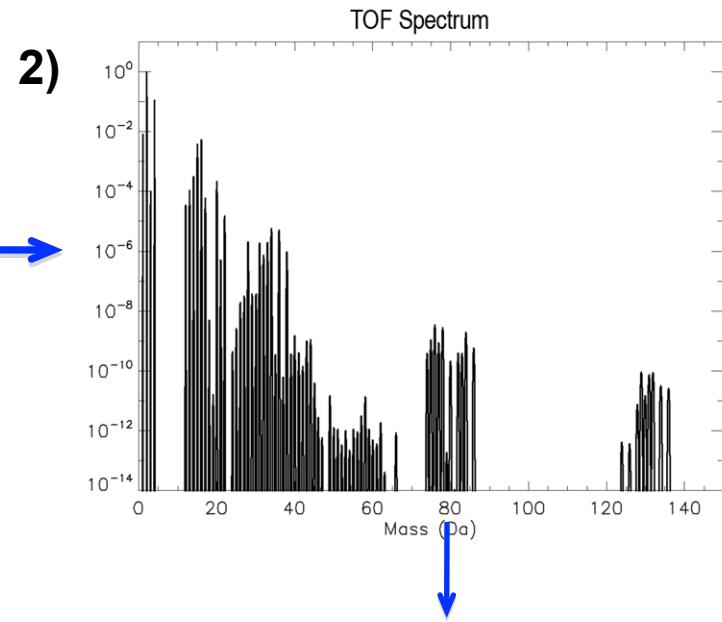
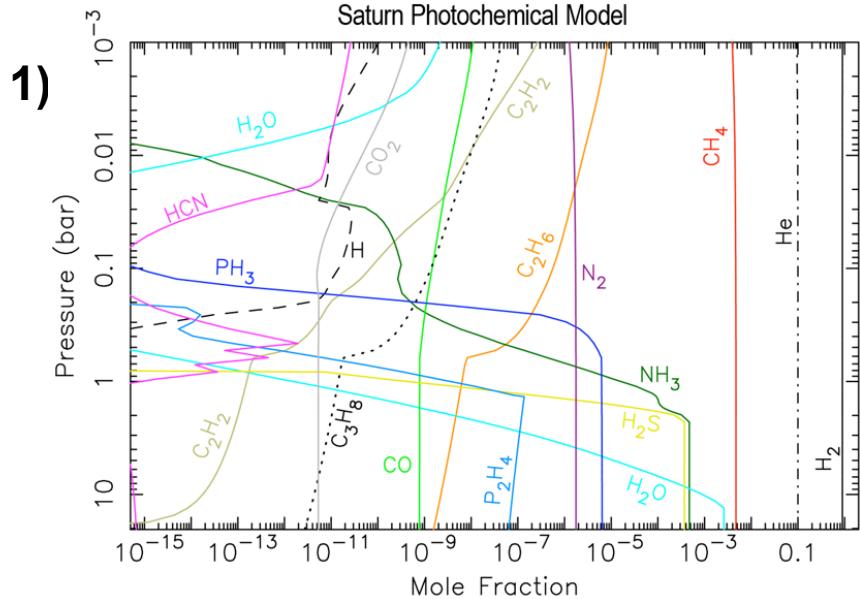


Third generation MBTOF has successfully undergone vacuum and vibration tests.

# MASPEX Performance (High-Resolution Mode)



# Scientific Requirements



**STEP 1:** Saturn atmosphere model (J. Moses) produced as the basis for estimating the mass spectrometric measurement requirements (over 100 compounds from 0.5 to 20 bars).

**STEP 2:** Simulated mass spectrum generated using empirical laboratory data (line shapes) from mass spectrometer combined with NIST fragmentation and ionisation data and solar isotopic abundance information (H. Waite).

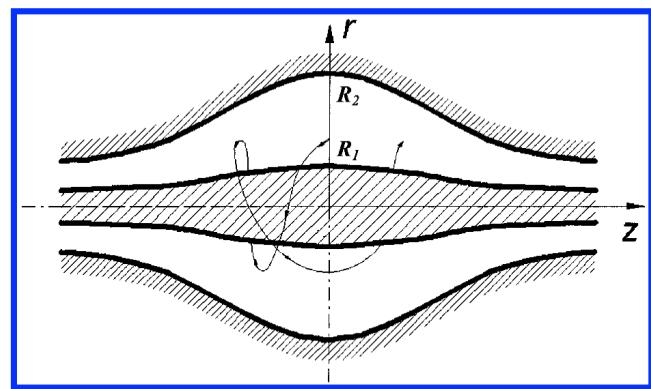
**STEP 3:** Spreadsheet programme developed to determine mass resolution and measurement time needed to satisfy the requirements and thus generate a realistic operational scenario (H. Waite).

3)

Results for the Measurement of Ambient Gas at 0.5 bar

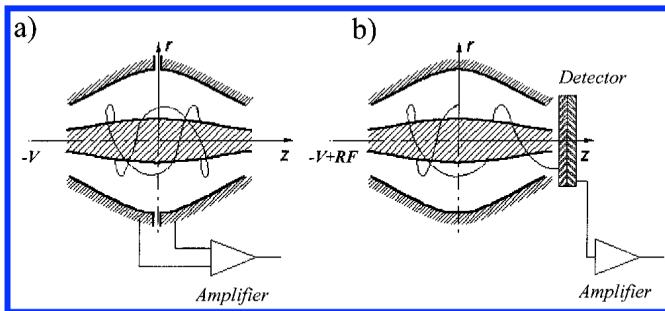
Molecule	Specific isotope	Exact mass RMM g/mol	Principal isotope	Molecular abundance	Mixing fraction	Min required acquisition time sec	Target precision	Minimum bounces required
H <sub>2</sub>	<sup>1</sup> H <sub>2</sub>	2.0	P	0.895	0.895	<0.1	5%	0
H <sub>2</sub>	<sup>1</sup> H <sup>1</sup> H	3.0	2H	0.895	3.58E-5	<0.1	5%	4
He	<sup>3</sup> He	3.0	3He	0.1	4.64E-5	<0.1	10%	4
He	<sup>4</sup> He	4.0	P	0.1	0.1	<0.1	5%	0
CH <sub>4</sub>	<sup>12</sup> C <sup>1</sup> H <sub>4</sub>	16.0	P	0.005	0.005	<0.1	1%	0
CH <sub>4</sub>	<sup>13</sup> C <sup>1</sup> H <sub>4</sub>	17.0	13C	0.005	5.25E-5	1.7	1%	0
CH <sub>4</sub>	<sup>14</sup> C <sup>1</sup> H <sub>4</sub>	17.0	2H	0.005	3.72E-7	9.7	5%	21
N <sub>2</sub>	<sup>14</sup> N <sub>2</sub>	28.0	P	1.75E-6	1.74E-6	0.3	10%	3
Ne	<sup>20</sup> Ne	20.0	P	2.06E-4	1.92E-4	<0.1	10%	0
Ne	<sup>22</sup> Ne	22.0	22Ne	2.06E-4	1.39E-5	0.2	10%	0
NH <sub>3</sub>	<sup>15</sup> NH <sub>3</sub>	17.0	P	1.39E-7	1.39E-7	7.7	10%	8
Ar	<sup>36</sup> Ar	36.0	P	5.37E-6	1.81E-8	21.1	10%	0
Ar	<sup>38</sup> Ar	38.0	38Ar	5.37E-6	3.41E-9	111.7	10%	0
C <sub>2</sub> H <sub>6</sub>	<sup>12</sup> C <sup>2</sup> H <sub>6</sub>	30.0	P	3.34E-8	3.27E-8	2219.5	1%	3
C <sub>2</sub> H <sub>6</sub>	<sup>13</sup> C <sup>2</sup> H <sub>6</sub>	31.1	13C	3.34E-8	7.38E-10	98205.	1%	16
Kr	<sup>79</sup> Kr	79.9	80Kr	3.04E-9	6.89E-11	13948.3	10%	3
Kr	<sup>81</sup> Kr	81.9	82Kr	3.04E-9	3.51E-10	2734.5	10%	0
Kr	<sup>82</sup> Kr	82.9	83Kr	3.04E-9	3.49E-10	2748.1	10%	0
Kr	<sup>83</sup> Kr	83.9	P	3.04E-9	1.74E-9	552.4	10%	0
Kr	<sup>85</sup> Kr	85.9	86Kr	3.04E-9	5.25E-10	1829	10%	0
Xe	<sup>127</sup> Xe	127.9	128Xe	3.04E-10	2.18E-12	229816.6	10%	0
Xe	<sup>128</sup> Xe	128.9	P	3.04E-10	2.21E-10	2269.6	10%	0
Xe	<sup>129</sup> Xe	129.9	130Xe	3.04E-10	4.61E-12	108669.6	10%	0
Xe	<sup>130</sup> Xe	130.9	131Xe	3.04E-10	2.41E-11	20797.4	10%	0
Xe	<sup>131</sup> Xe	131.9	132Xe	3.04E-10	3.05E-11	16409.1	10%	0
Xe	<sup>133</sup> Xe	133.9	134Xe	3.04E-10	1.19E-11	42291.8	10%	0
Xe	<sup>135</sup> Xe	135.9	136Xe	3.04E-10	1.E-11	49875.7	10%	0

# Orbitrap



**Figure 1.** Equipotentials of the quadro-logarithmic field and an example of a stable ion trajectory

$$U(r,z) = \frac{k}{2} \left( z^2 - \frac{r^2}{2} \right) + \frac{k}{2} (R_m)^2 \ln \left[ \frac{r}{R_m} \right] + C$$



**Figure 2.** Modes of mass analysis in the orbitrap: (a) Fourier transform mass spectrometry (image current detection); (b) mass-selective instability (detection using secondary electron multiplier).

## ❖ Early work

- A. Makarov, Electrostatic axially harmonic orbital trapping: a highperformance technique of mass analysis. *Anal. Chem.* 72, (2000) 1156–1162
- Qizhi Hu, Robert J. Noll, Hongyan Li, Alexander Makarov, Mark Hardman, and R. Graham Cooks, The Orbitrap: a new mass spectrometer, *J. Mass Spectrom.* 2005; 40: 430–443

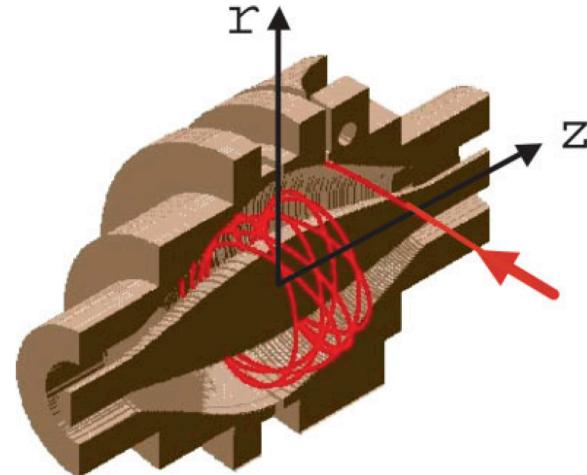
## ❖ Theory

- Andriy Kharchenko, Gleb Vladimirov, Ron M. A. Heeren, Eugene N. Nikolaev, Performance of Orbitrap Mass Analyzer at Various Space Charge and Non-Ideal Field Conditions: Simulation Approach, *J. Am. Soc. Mass Spectrom.* (2012) 23:977–987

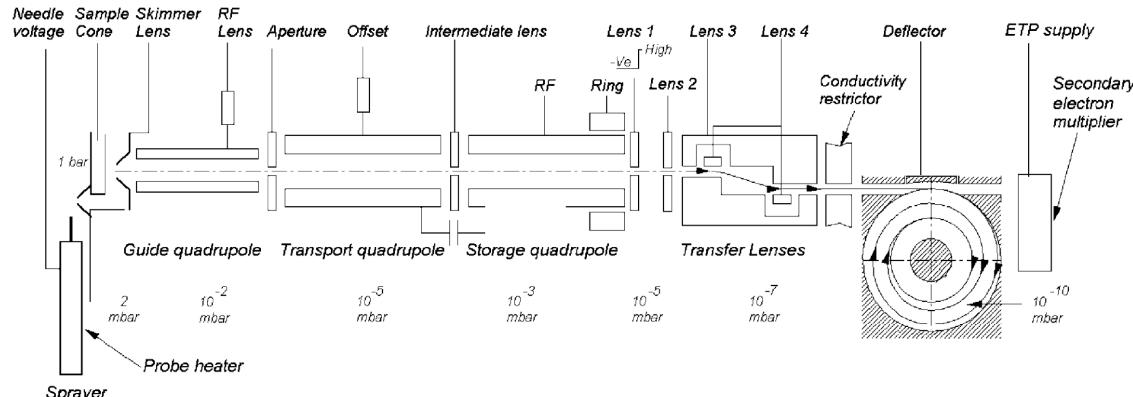
## ❖ Multiple MS ( $MS^M$ )

- Robert Cho, Yingying Huang, Jae C. Schwartz, Yan Chen, Timothy J. Carlson, Ji Ma,  $MS^M$ , an Efficient Workflow for Metabolite Identification Using Hybrid Linear Ion Trap Orbitrap Mass Spectrometer, *J. Am. Soc. Mass Spectrom.* (2012) 23:880–888

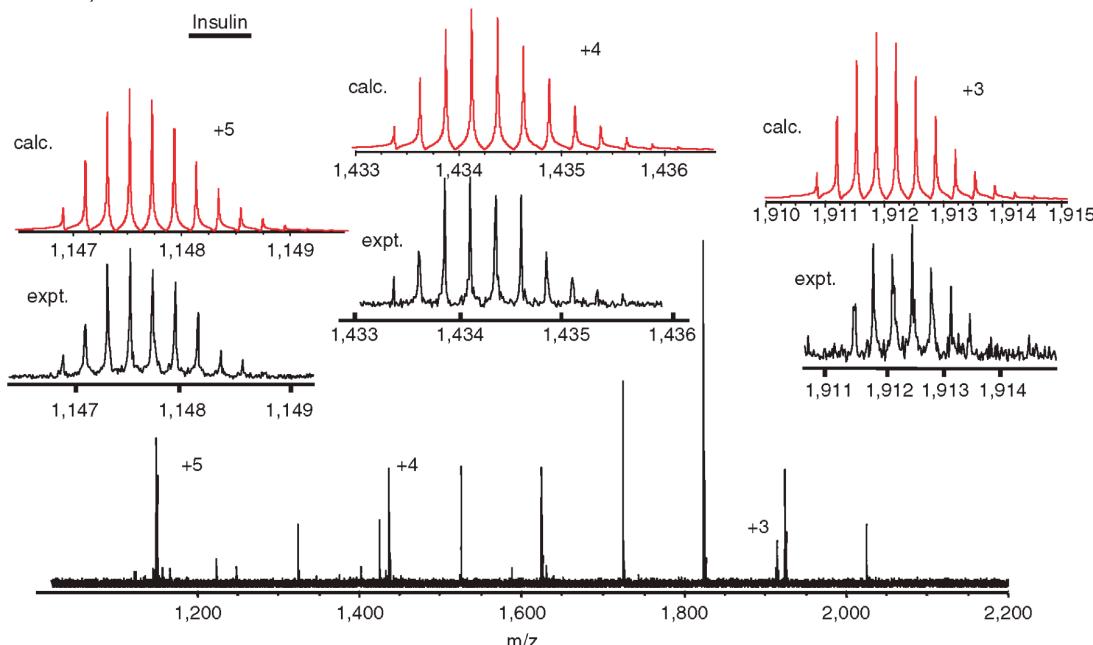
# Orbitrap



**Figure 1.** Cutaway view of the Orbitrap mass analyzer. Ions are injected into the Orbitrap at the point indicated by the red arrow. The ions are injected with a velocity perpendicular to the long axis of the Orbitrap (the  $z$ -axis). Injection at a point displaced from  $z = 0$  gives the ions potential energy in the  $z$ -direction. Ion injection at this point on the  $z$ -potential is analogous to pulling back a pendulum bob and then releasing it to oscillate.



**Figure 2.** The experimental Orbitrap mass spectrometer. Ions are produced by the electrospray ion source at the extreme left. Ions then proceed through the source, collision quadrupole, selection quadrupole and then pass into the storage quadrupole. The storage quadrupole serves as an ion accumulator and buncher, allowing a pulsed mass analyzer such as the Orbitrap to be coupled to a continuous source like an electrospray ionization source. After accumulation and bunching in the storage quadrupole, the exit lens ('Lens 1') is pulsed low, the ion bunches traverse the ion transfer lens system and are injected into the Orbitrap mass analyzer (shown end-on).



**Figure 3.** ESI mass spectrum of bovine insulin. Data acquisition parameters include a data sampling rate of 5 MHz, record length was 8 million data points, and the Fourier transform was performed with no apodization function or zero-filling. The lower spectrum shows a wide range mass spectrum including the internal mass calibrant Ultramark 1621 whose oligomers are spaced by 100 mass/charge unit intervals. Lower traces in the close-ups show experimentally obtained isotopic distributions for each charge state. Upper traces in the close-ups show the theoretically expected isotopic distributions. The calculated isotope distributions were obtained from IsoPro 3.0 using Gaussian peak shapes with resolution of 100 000.

Qizhi Hu, Robert J. Noll, Hongyan Li, Alexander Makarov, Mark Hardman, and Graham R. Cooks, The Orbitrap: a new mass spectrometer, *J. Mass Spectrom.* 2005; 40: 430–443

# Reduced instrument resources

# Micro-Gas Chromatography Systems

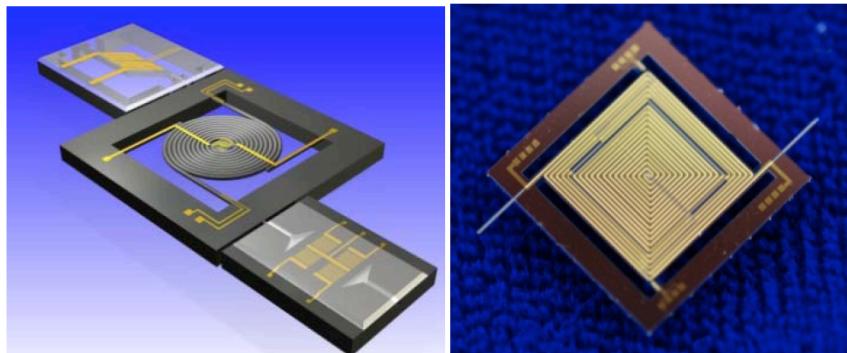


Figure 1: Left: Drawing of complete  $\mu$ GC analysis system. Right: A 25cm-long separation column with built-in fluidic interconnects.

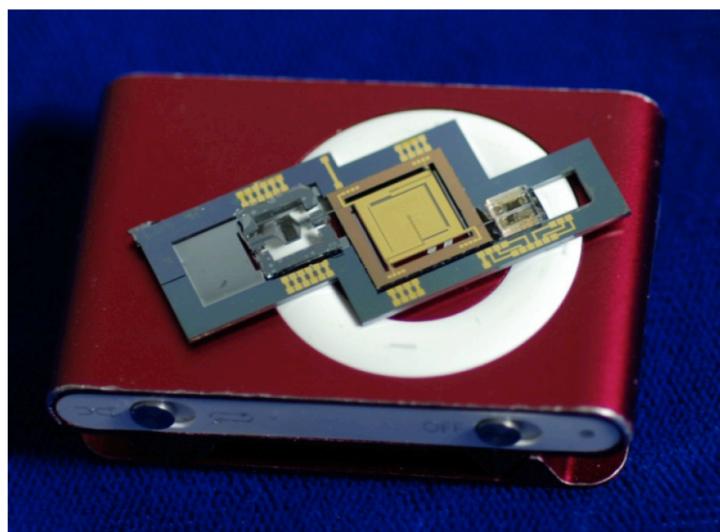


Figure 9: A 25cm separation column, single-bed preconcentrator, and chemi-resistive detector on a 2<sup>nd</sup> generation Apple iPod Shuffle about the size of the intended  $\mu$ GC system. The outer circle is about the size of a U.S. quarter.

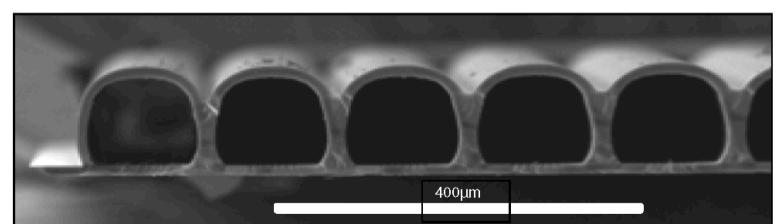
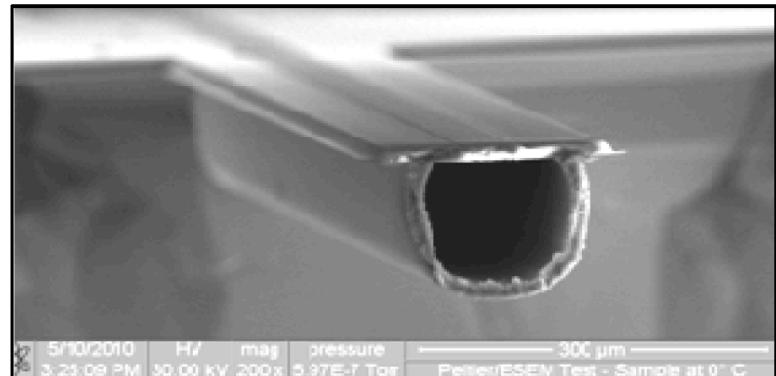


Figure 3: Cross-sections of the laser-cut fluidic interconnects and the column cross-section. The flow channel is 120  $\mu$ m across and 90  $\mu$ m deep.

K.T.M. Beach, S.M. Reidy, R.J.M. Gordenker, and K.D. Wise, A low-mass high-speed  $\mu$ GC separation column with built-in fluidic chip-to-chip interconnects, IEEE proc. 2011

# Micro-Gas Chromatography Systems (GC-GC)

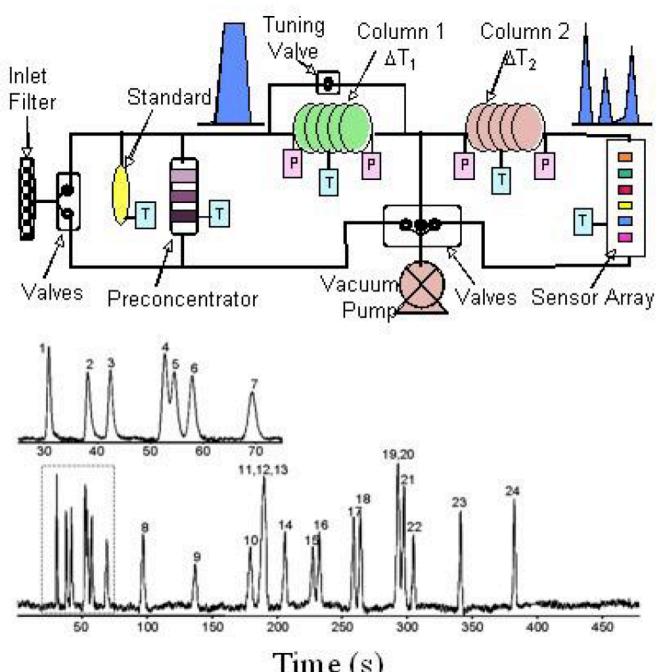


Fig. 1: Block diagram of a microfabricated chromatography-based gas analyzer (above), with a chromatogram on the separation of air-phase petroleum hydrocarbons.

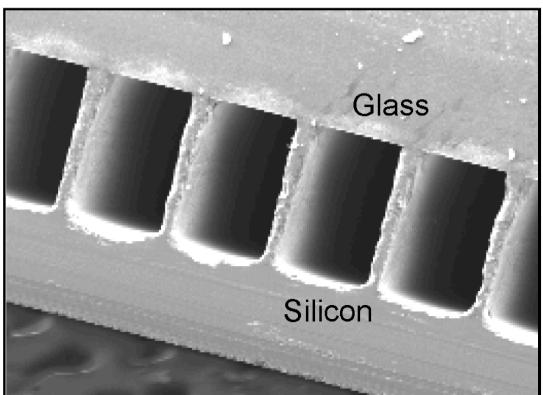
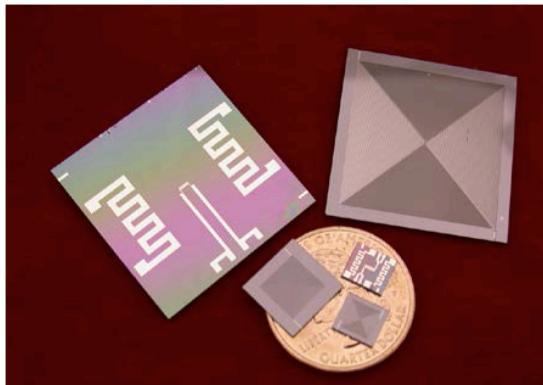


Fig. 4: 2D-GC columns (3m, 25cm, and 10cm) having optimized dimensions on a U.S. quarter, with heaters and sensors to allow closed-loop temperature program on their back surfaces.

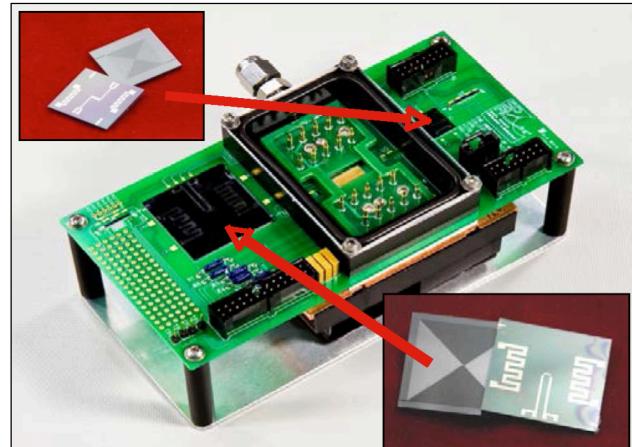


Fig. 6: The GCxGC system, with insets of the first-dimension column (bottom right) and the second-dimension column (top left).

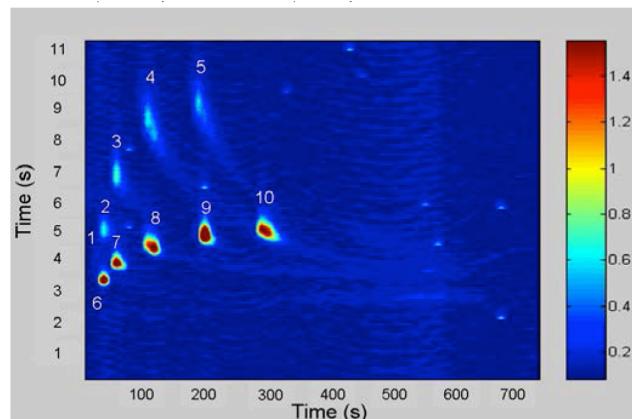


Fig. 7: A two-dimensional contour plot of a 10-component GCxGC separation using 3m- and 0.25m-long microcolumns and a two-stage thermal modulator. Alkanes are separated primarily by the first column (x-axis) and the ketones are separated primarily by the second column (y-axis). Color reflects peak height on the FID.

S. Reidy, S.-J. Kim, K. Beach, B. Block, E.T. Zellers, K. Kurabayashi, and K.D. Wise, A micro-fabricated two-dimensional gas chromatography system, 2010

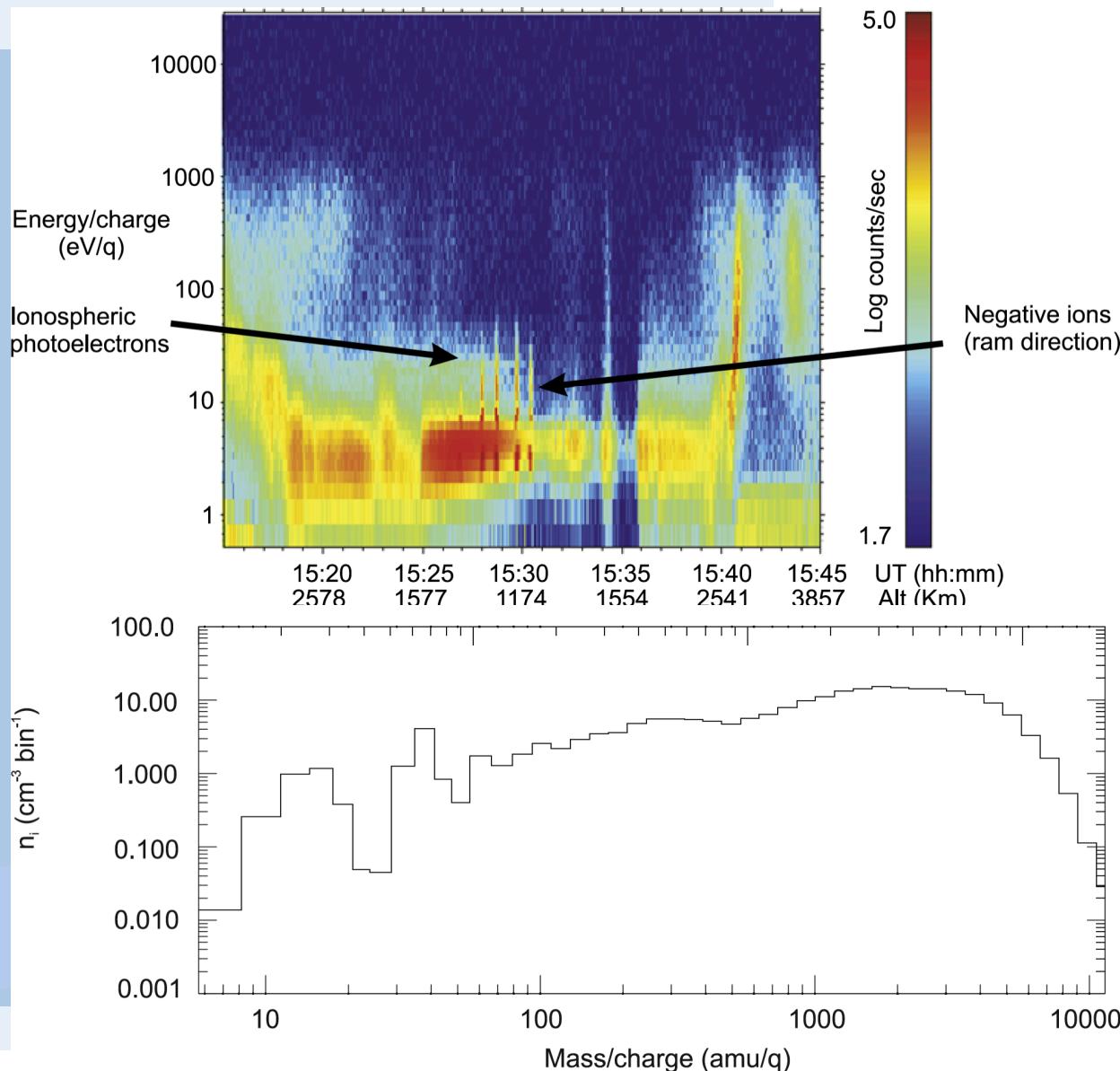
# Summary

## Possible Future Developments

- ❖ Increase in sensitivity
  - Non-scanning instruments
    - Time of flight MS
    - Cryo trap
- ❖ Increased mass range
  - Time-of-flight MS
  - Orbitrap MS
- ❖ Increase in mass resolution
  - ROSINA instrument package
    - DFMS: Double focusing mass spectrometer
    - RTOF: Time-of-flight mass spectrometer
  - Multi-bounce TOF
  - Orbitrap MS
- ❖ Reduced complexity
- ❖ Reduced instrument resources
  - Micro-GC systems
  - TOF
  - Orbitrap

# Negative Ions in Titan's Upper Atmosphere

- ❖ Negative ions in the upper atmosphere
- ❖ Typical mass groups
  - M: 10–30, 30–50, 50–80, 80–110, and 110–200 amu
- ❖ Ions with very high masses
  - M < 10'000 amu
- ❖ Important for the formation of organic-rich aerosols (tholins) eventually falling to the surface



A. Coates et al., Geophys.  
Res. Lett. 34 (2007) L22103

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# Reduced Complexity

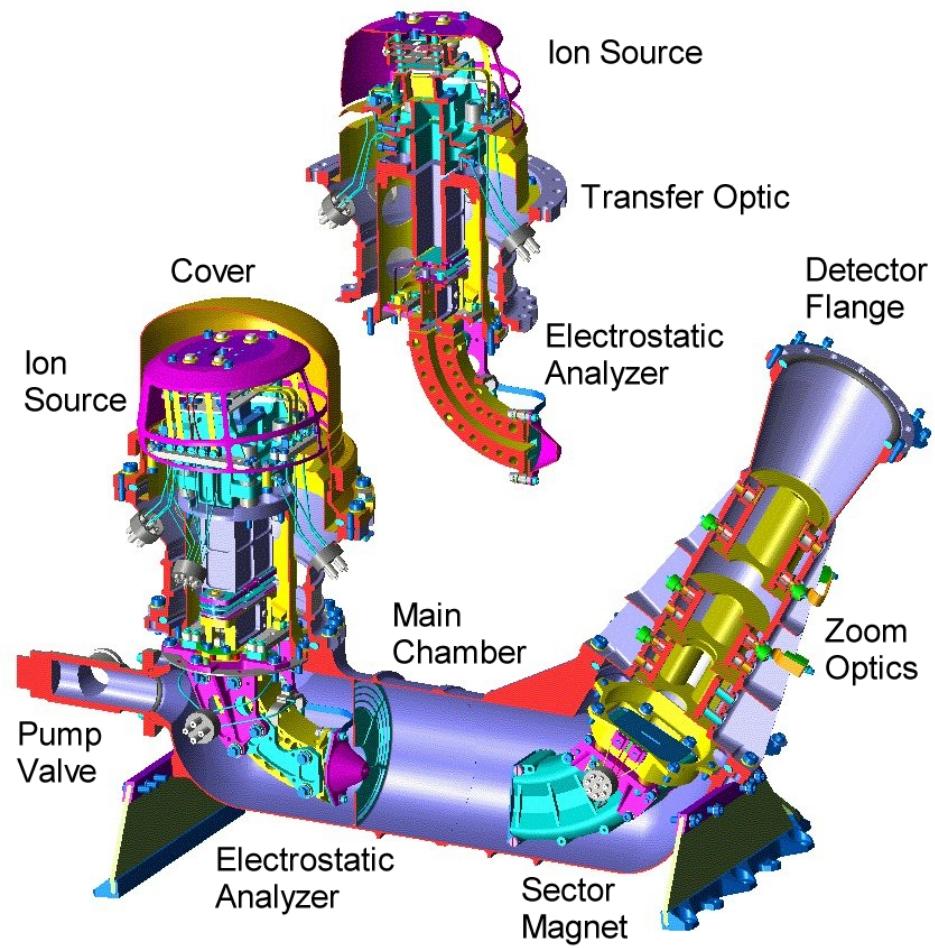
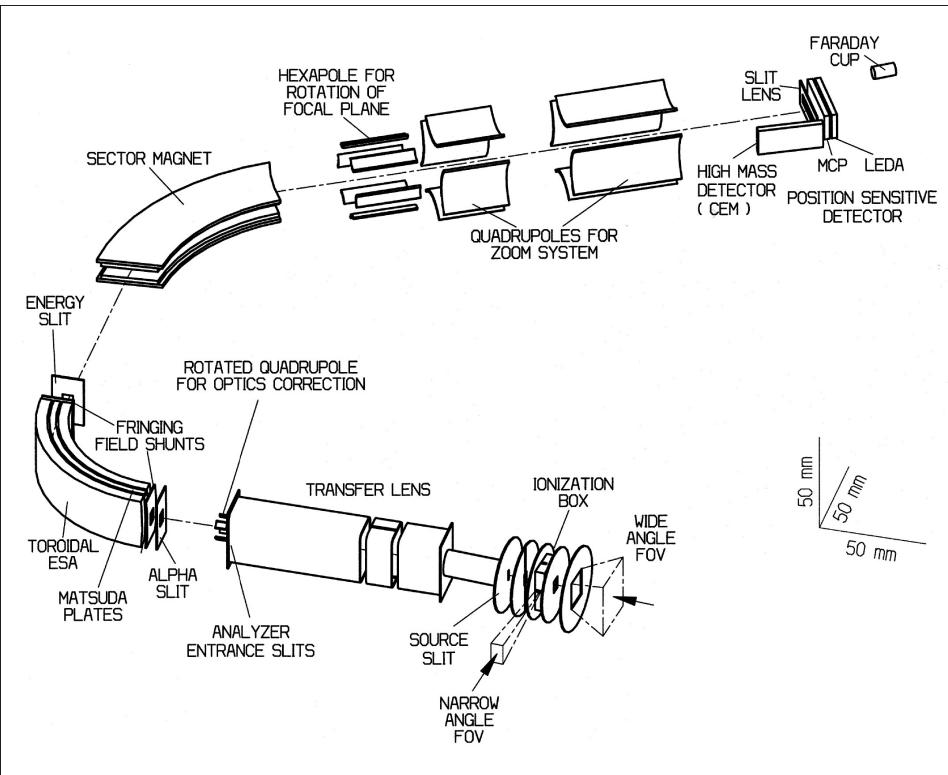
*u*<sup>b</sup>

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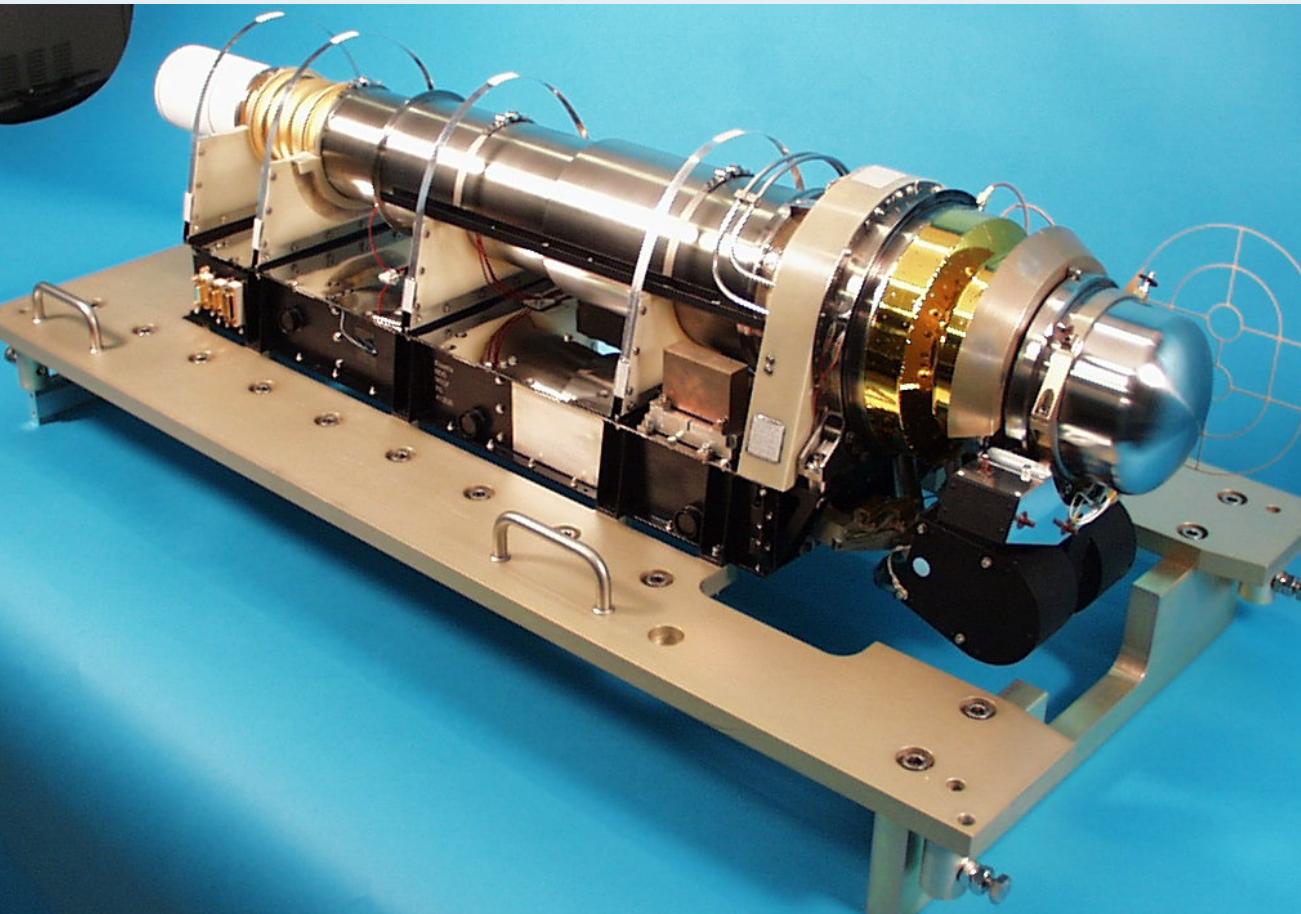
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# Reduced Complexity

# Rosetta / ROSINA / DFMS Sensor



# RTOF / ROSINA on Rosetta Mission

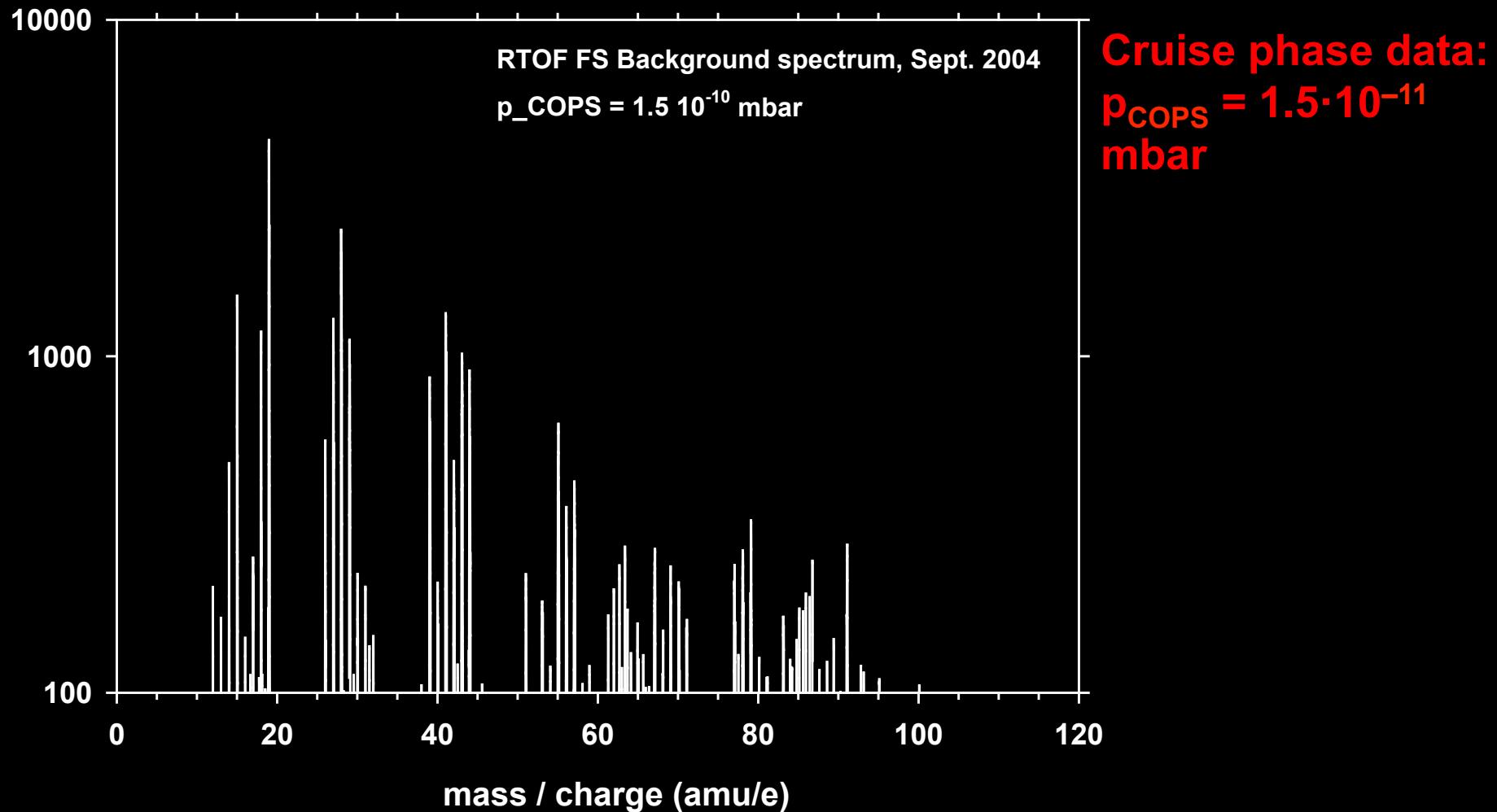


- ❖ RTOF design: 1996
- ❖ Mass resolution:  $m/\Delta m = 1500 \dots 4500$  (depending on mode)
- ❖ Mass range: 1 ... 500 amu (in principle unlimited, because TOF instrument)
- ❖ Sensitivity:  $10^{-4} \dots 10^{-3}$  A/mbar (depending on mode)
- ❖ Mass: 14.7 kg
- ❖ Power: 28 W

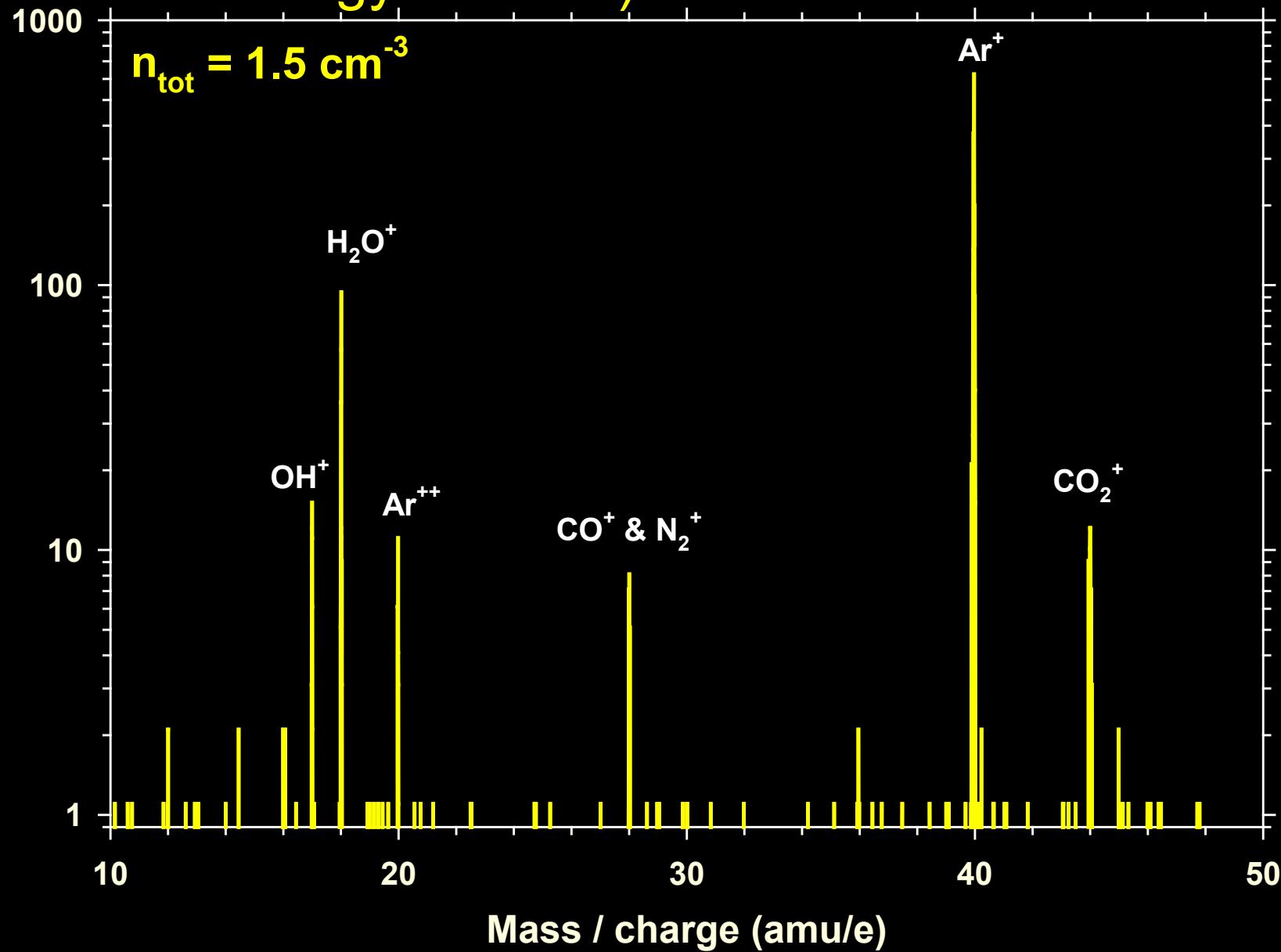
H. Balsiger, et al., "ROSINA - Rosetta Orbiter Spectrometer for Ion and Neutral Analysis," Space Science Review 128 (2007), 745–801.

S. Scherer, K. Altwegg, H. Balsiger, J. Fischer, A. Jäckel, A. Korth, M. Mildner, D. Piazza, H. Rème, and P. Wurz, A novel principle for an ion mirror design in time-of-flight mass spectrometry, Int. Jou. Mass Spectr. 251 (2006) 73–81.

# RTOF / ROSINA on Rosetta mission

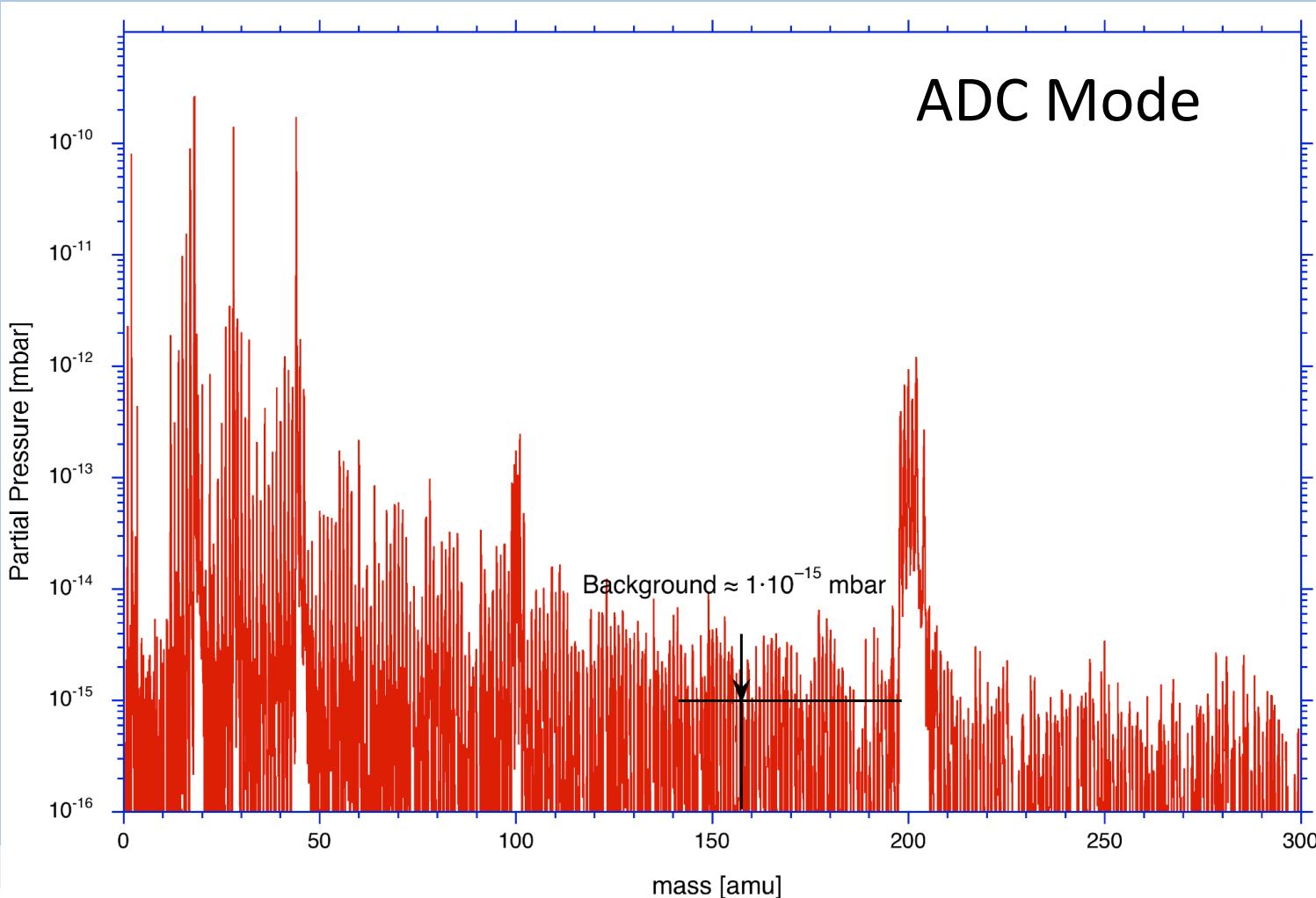


# RTOF/ ROSINA ion spectrum Ion energy < 15 eV)



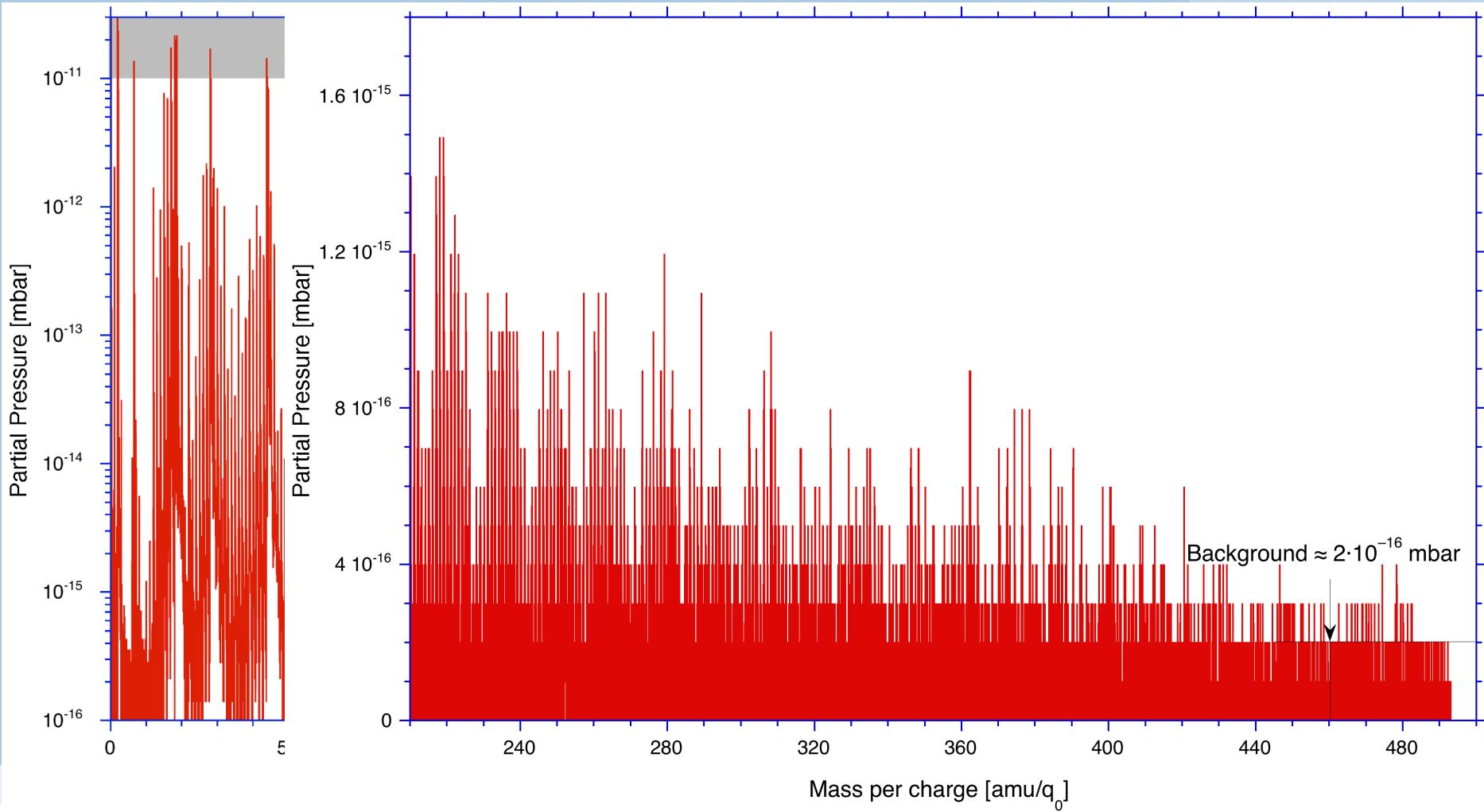
# Prototype Results: Sensitivity

NGMS Sensitivity, static, ADC Mode



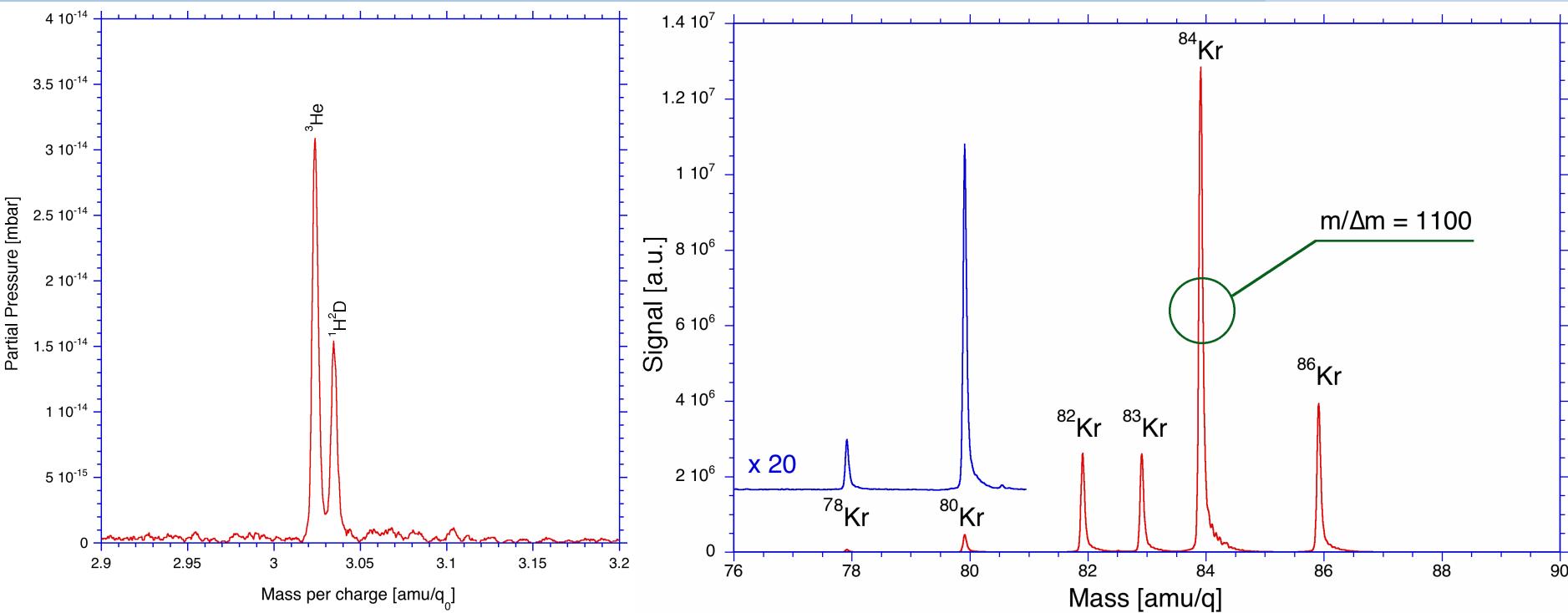
# Prototype Results: Sensitivity

## NGMS Sensitivity, static, TDC Mode



# Prototype Results: Mass Resolution

## Neutral Gas mass Spectrometer (NGMS)

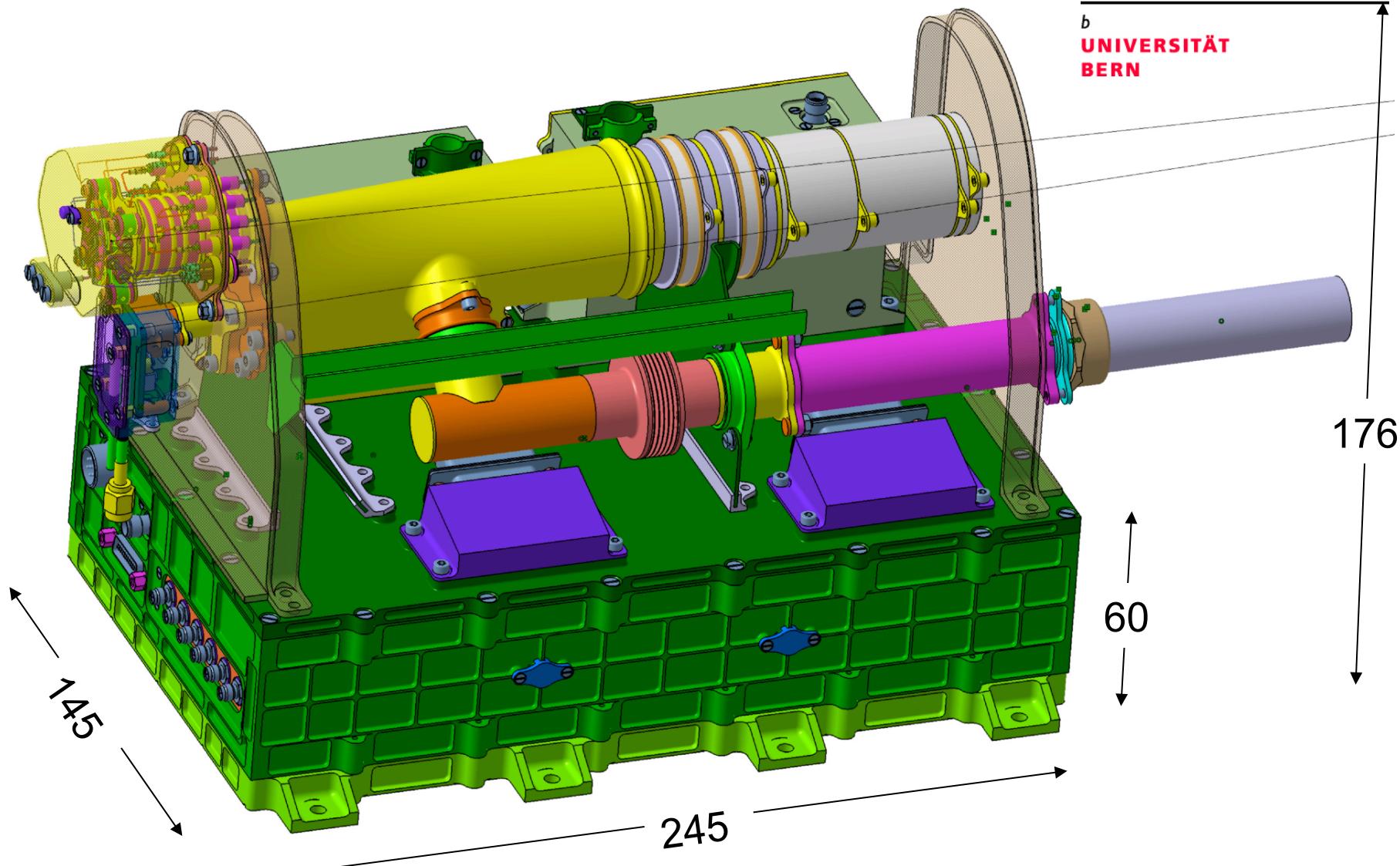


P. Wurz, D. Abplanalp, M. Tulej, and H. Lammer, "A Neutral Gas Mass Spectrometer for the Investigation of Lunar Volatiles," *Planet. Sp. Science* (2012) in print.

# Neutral Gas Mass Spectrometer Luna-Resurs / Luna-Glob

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P. Wurz, D. Abplanalp, M. Tulej, and H. Lammer, "A Neutral Gas Mass Spectrometer for the Investigation of Lunar Volatiles," *Planet. Sp. Science* (2012) in print.

# Summary

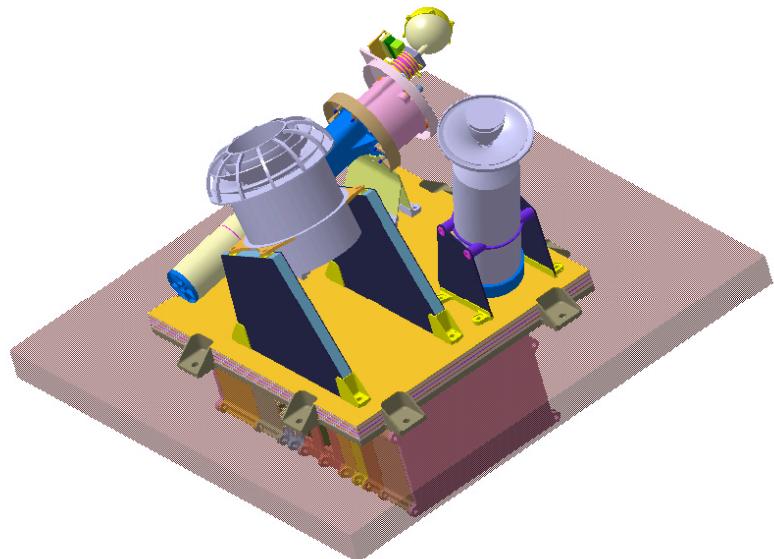
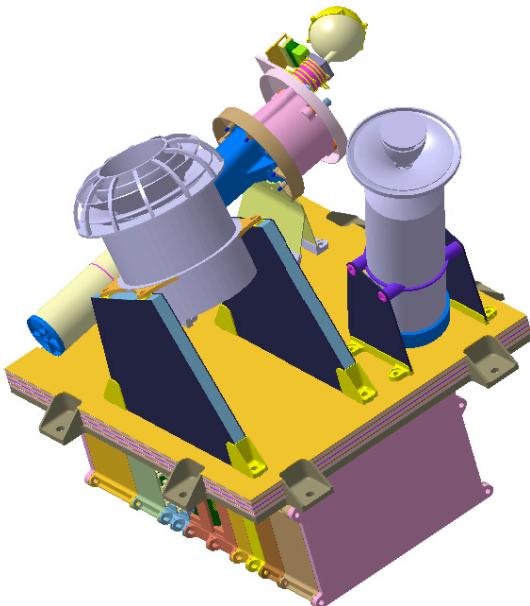
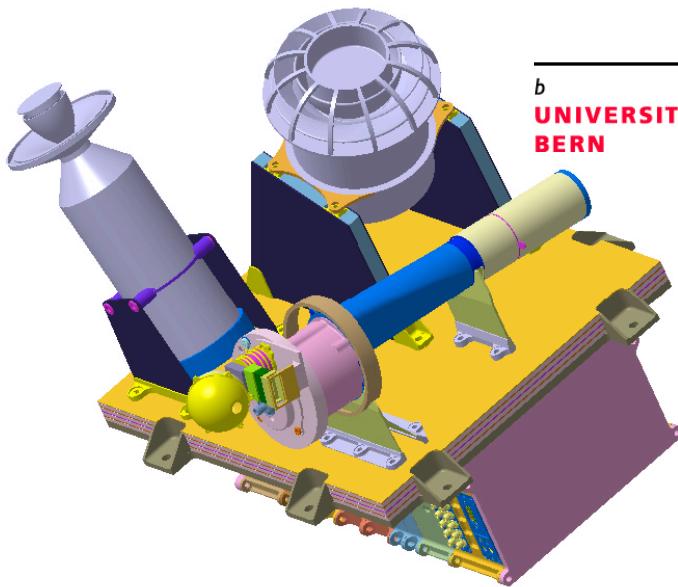
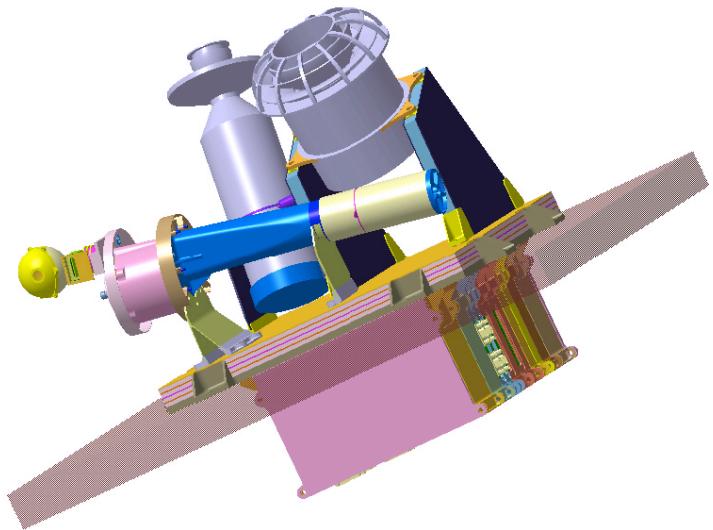
- ❖ We developed a mass spectrometer for gas analysis of atmospheres and exospheres
  - Mass range 1 – 1000 amu
  - Sensitivity  $\approx 1 \text{ cm}^{-3}$  in 10 s integration
  - Prototype successfully used for stratospheric research
    - Abplanalp et al. Adv. Sp. Res. 2010, Wieser et al. Adv. Sp. Res. 2010
  - Flight design for Luna-Resurs
    - Mass 3.5 kg, power 17W
- ❖ Luna-Resurs: GC-MS
  - Investigation of the volatiles contained in the soil by GC-MS analysis
  - Investigation of the lunar exosphere
    - Contamination by spacecraft
    - The Rosetta experience: Schläppi et al. JGR 2010

# PEP-NU Design

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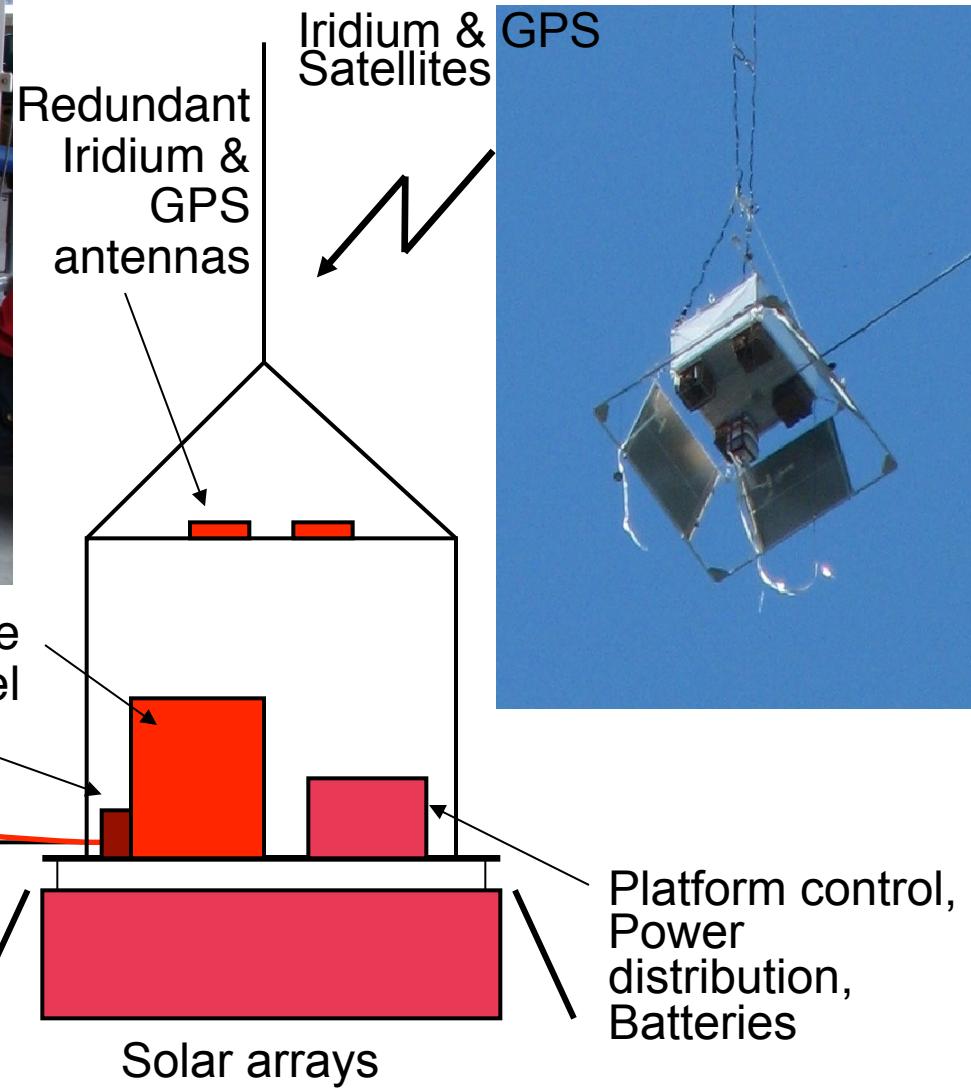
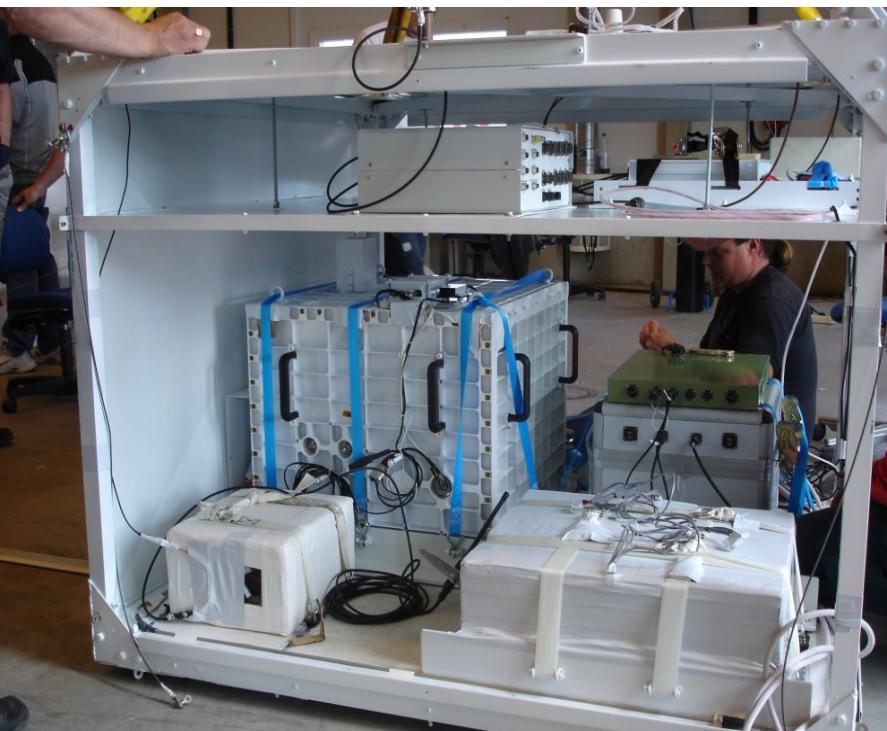
# Summary

- ❖ PEP-NU Design overview
  - NIM progress since PM #5
    - Luna-Resurs / Luna-Globe design
    - Prototype measurements
  - PEP-NU no progress
    - Wait for input on LEN and ELS
    - Decide on instrument placement
- ❖ Radiation shielding
  - Detailed radiation analysis for PEP-NU on JGO
    - JGO-PEP-TN-0901-I1R0 (Radiation Analysis PEP-NU).pdf
  - Total dose increase
    - From 38 to 50 krad Si behind 20 mm Al equivalent.
  - Assessment of Europa flybys

*u*<sup>b</sup>

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# P-BACE on the MEAP platform

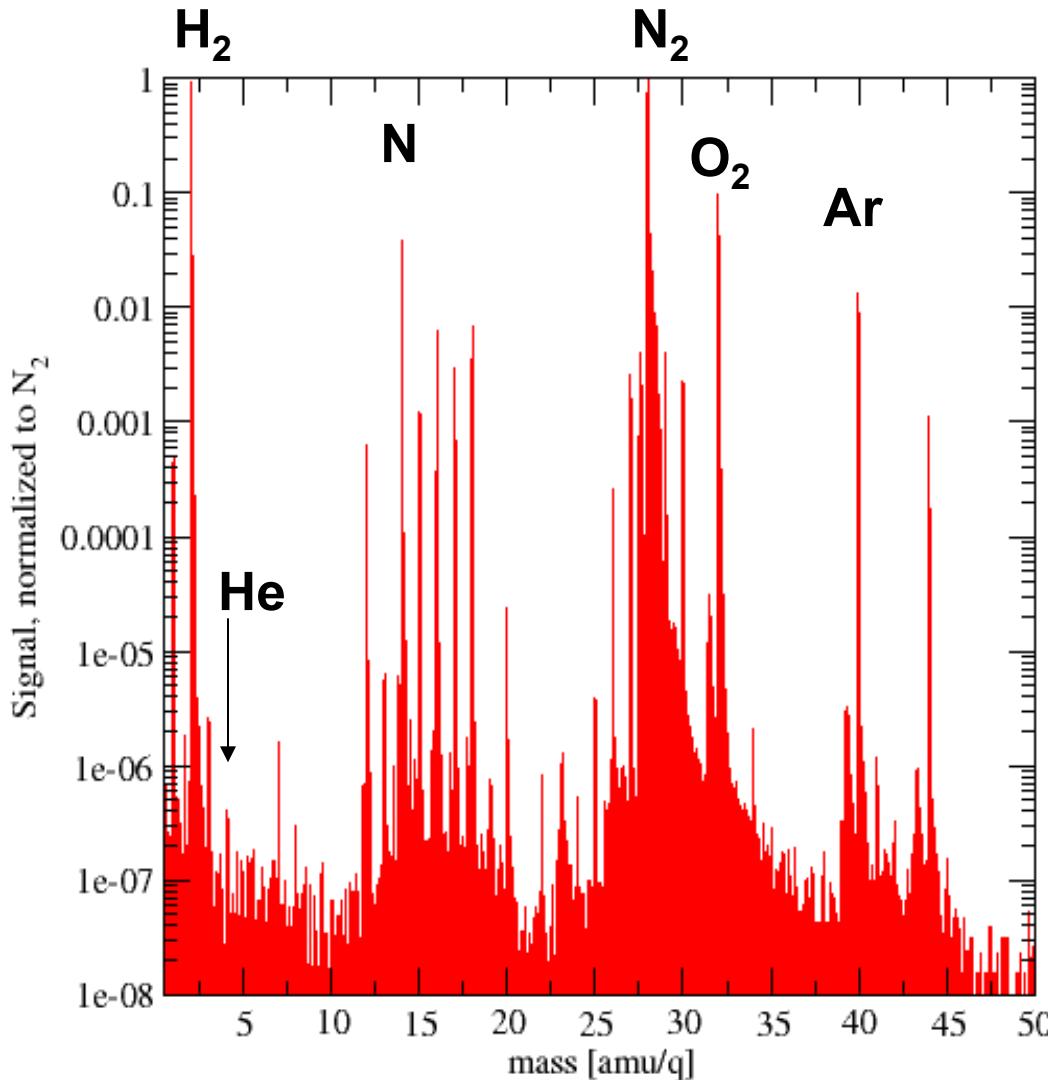


# MEAP flight path

- ❖ Balloon provided by Esrange Space Center to test beyond line of sight flight
- ❖ Semicircular flight following the summer polar vortex
- ❖ Launched from Esrange, Sweden on 28 June 2008
- ❖ 116 hours flight time
- ❖ Altitude 33 ... 38 km
- ❖ Landed in Canada, near Umingmaktok, on 3 July 2008.
- ❖ Recorded ~ 4500 mass spectra in stratosphere



# P-BACE quicklook data, dynamic range



- ❖ Raw data
- ❖ No background subtracted
- ❖ Dynamic range:  
6–7 orders of magnitude per spectrum
- ❖ Mass range:  
1–1000 amu/q
- ❖ D. Abplanalp, P. Wurz, L. Huber, I. Leya, E. Kopp, U. Rohner, M. Wieser, L. Kalla, and S. Barabash, Adv. Space Res. 44 (2009) 870–878.